

## Exploring Microstructure and Bending Strength of Al<sub>2</sub>O<sub>3</sub> Ceramics Doped with Sm<sub>2</sub>O<sub>3</sub> Rare-Earth Oxide: Impact of Volume Ratios and Sintering Temperatures

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### Abstract

The effects of low amounts (0, 0.1, 0.3, 0.5, 0.8, 1, 2 vol%) of Sm<sub>2</sub>O<sub>3</sub> on the densification, microstructure, and bending strength of Al<sub>2</sub>O<sub>3</sub> were studied for different sintering temperatures. Sm<sub>2</sub>O<sub>3</sub> and Al<sub>2</sub>O<sub>3</sub> powders were mixed by ball milling, and Al<sub>2</sub>O<sub>3</sub>-Sm<sub>2</sub>O<sub>3</sub> ceramics were prepared by dry and cold isostatic pressing in disc form before pressureless sintering at 1550°C and 1600°C/2 h in the air, separately. Sm<sub>2</sub>O<sub>3</sub> reacted with Al<sub>2</sub>O<sub>3</sub> by forming SmAlO<sub>3</sub>. The rod-like morphology of the SmAlO<sub>3</sub> was generally achieved for 0.8 and 1 vol% Sm<sub>2</sub>O<sub>3</sub> for sintering at 1600°C, whereas the rod-like form was more obvious for the samples sintered at 1550°C. Similar relative densities were attained for all Sm<sub>2</sub>O<sub>3</sub> ratios at both sintering temperatures, but the Al<sub>2</sub>O<sub>3</sub> sintered at 1550°C exhibited higher densification than the Al<sub>2</sub>O<sub>3</sub> sintered at 1600°C. The strength values were close, while a strength increase of about 5% was obtained for 0.5 vol% Sm<sub>2</sub>O<sub>3</sub> containing ceramics sintered at 1600°C caused by the higher densification compared to the Al<sub>2</sub>O<sub>3</sub>. The strength showed a prominent drop above the ratio of 0.1 vol% Sm<sub>2</sub>O<sub>3</sub> for sintering at 1550°C. The addition of present amounts of Sm<sub>2</sub>O<sub>3</sub> did not have a significant effect on the mechanical properties of Al<sub>2</sub>O<sub>3</sub> but it changed the microstructure.

**Keywords:** Al<sub>2</sub>O<sub>3</sub>, Sm<sub>2</sub>O<sub>3</sub>, microstructure, mechanical properties.

## Sm<sub>2</sub>O<sub>3</sub> Nadir Toprak Oksit Katkılı Al<sub>2</sub>O<sub>3</sub> Seramiklerinin Mikro Yapısının ve Eğilme Dayanımının İncelenmesi: Hacim Oranlarının ve Sinterleme Sıcaklıklarının Etkisi

### Öz

Düşük miktarlarda (0, 0,1, 0,3, 0,5, 0,8, 1, 2 hac.%) Sm<sub>2</sub>O<sub>3</sub>'ün Al<sub>2</sub>O<sub>3</sub>'ün yoğunlaşma, mikroyapı ve eğilme dayanımı üzerindeki etkileri farklı sinterleme sıcaklıkları için incelenmiştir. Sm<sub>2</sub>O<sub>3</sub> ve Al<sub>2</sub>O<sub>3</sub> tozları bilyeli öğütme ile karıştırılmış ve Al<sub>2</sub>O<sub>3</sub>-Sm<sub>2</sub>O<sub>3</sub> seramikleri, ayrı ayrı 1550°C'de ve 1600°C/2 saat havada basınçsız sinterlemeden önce disk formunda kuru ve soğuk izostatik presleme ile hazırlanmıştır. Sm<sub>2</sub>O<sub>3</sub>, Al<sub>2</sub>O<sub>3</sub> ile reaksiyona girerek SmAlO<sub>3</sub> oluşturmuştur. SmAlO<sub>3</sub>'ün çubuk benzeri morfolojisi, 1600°C'de sinterleme için genel olarak hacimce %0,8 ve 1 Sm<sub>2</sub>O<sub>3</sub> için elde edilirken, 1550°C'de sinterlenen numunelerde çubuk benzeri form daha belirgin olmuştur. Her iki sinterleme sıcaklığında da tüm Sm<sub>2</sub>O<sub>3</sub> oranları için benzer relatif yoğunluklar elde edilmiştir ancak 1550°C'de sinterlenen Al<sub>2</sub>O<sub>3</sub>, 1600°C'de sinterlenen Al<sub>2</sub>O<sub>3</sub>'e göre daha yüksek yoğunlaşma sergilemiştir. 1600°C'de sinterlenen hac.%0,5 Sm<sub>2</sub>O<sub>3</sub> içeren seramiklerde, Al<sub>2</sub>O<sub>3</sub>'e göre daha yüksek yoğunlaşma sebebiyle yaklaşık %5'lik bir mukavemet artışı elde edilirken, mukavemet değerleri genelde birbirine yakın olmuştur. Mukavemet, 1550°C'de sinterleme için hac.%0,1 Sm<sub>2</sub>O<sub>3</sub> oranının üzerinde belirgin bir düşüş göstermiştir. Mevcut miktarlarda Sm<sub>2</sub>O<sub>3</sub> ilavesinin Al<sub>2</sub>O<sub>3</sub>'ün mekanik özellikleri üzerinde önemli bir etkisi olmamıştır, ancak mikroyapıyı değiştirmiştir.

**Anahtar Kelimeler:** Al<sub>2</sub>O<sub>3</sub>, Sm<sub>2</sub>O<sub>3</sub>, mikroyapı, mekanik özellikler.

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## 1. Introduction

$\text{Al}_2\text{O}_3$  ceramics can be counted as a classic representative of advanced ceramics. The physical properties of these ceramics are mostly desirable for structural applications such as in defense industries, aerospace, and biomedical, mainly while the environmental conditions are specifically severe (Shuai et al., 2020; Dresch et al., 2021; Ma et al., 2019). Nevertheless, the fracture toughness and flexural strength of  $\text{Al}_2\text{O}_3$  ceramics are low among the most commonly used technical ceramics like  $\text{Si}_3\text{N}_4$  and  $\text{SiC}$  (Hazell, 2016; Dresch et al., 2021; Flinders et al., 2005). This can be seen as a disadvantage when it comes to the use of these materials in critical applications (Huang and Chen, 2016; Kafkaslıoğlu Yıldız and Tür, 2021). The additives are greatly significant for the achievement of the sintering and, hence, for the enhancement of the mechanical properties (Lartigue-Korinek et al., 2002; Shi et al., 2020; Harun et al., 2012; Aktas et al., 2016). Latterly, various rare earth element oxides have been commonly used as sintering additives for  $\text{Al}_2\text{O}_3$  ceramics to enhance the densification process and improve mechanical properties (Rani et al., 2004). By using rare earth oxides, it is probable to decrease the sintering temperature of ceramics and enhance their microstructures. Rare earth oxides including  $\text{Sm}_2\text{O}_3$ ,  $\text{La}_2\text{O}_3$ ,  $\text{Dy}_2\text{O}_3$ ,  $\text{Y}_2\text{O}_3$ ,  $\text{CeO}_2$ ,  $\text{Er}_2\text{O}_3$ , and  $\text{Eu}_2\text{O}_3$  are commonly employed (Shi et al., 2020; Yijun et al., 2010; Ge et al., 2019; Aktas, Tekeli, Kucuktuvek, 2014; Aktas, Tekeli, Salman, 2014). Samarium (III) oxide ( $\text{Sm}_2\text{O}_3$ ) is a light rare earth element oxide with a melting temperature of over  $2000^\circ\text{C}$ , making it a suitable additive for the production of ceramics with enhanced sintering properties (Shi et al., 2020).  $\text{Sm}_2\text{O}_3$  rare earth oxides react with  $\text{Al}_2\text{O}_3$  by forming the  $\text{SmAlO}_3$  phase at high sintering temperatures, consistent with their phase diagram (Mizuno et al., 1977). In  $\text{Al}_2\text{O}_3$ - $\text{Sm}_2\text{O}_3$  ceramic systems, the formed  $\text{SmAlO}_3$  phase elongates along the growth direction, and corresponding  $\text{Al}_2\text{O}_3$ - $\text{SmAlO}_3$  ceramic composites have high mechanical properties (toughness, flexural strength) since this phase formed can act as short fiber reinforcement (Shi et al., 2020).

Shi et al. studied 20 vol.% Ti and (0-2.0 at%)  $\text{SmAlO}_3$  containing  $\text{Al}_2\text{O}_3$  ceramics in their study and obtained high toughness and bending strength values as a result of strengthening and toughening effects with the support of  $\text{SmAlO}_3$  phase (Shi et al., 2020). Also, Yijun et al. studied the effect of different rare earth oxides ( $\text{Y}_2\text{O}_3$ ,  $\text{La}_2\text{O}_3$ , and  $\text{Sm}_2\text{O}_3$ ) on the mechanical properties of  $\text{Al}_2\text{O}_3$  ceramics. The addition of rare-earth oxides enhanced the mechanical properties of  $\text{Al}_2\text{O}_3$  especially for 0.5 wt%  $\text{Sm}_2\text{O}_3$  content by way of grain size refinement and grain boundary strengthening mechanisms (Yijun et al., 2010). There are only a few studies in the literature examining the mechanical properties of  $\text{Sm}_2\text{O}_3$  doped  $\text{Al}_2\text{O}_3$  ceramics and the studies are mainly related to the creep and mechanical behavior of the eutectic composition (26 %  $\text{Sm}_2\text{O}_3$ + 74%  $\text{Al}_2\text{O}_3$ ) (Londaitzbehere et al., 2017; Ma et al., 2019). Therefore, in this work, the effects of different low amounts (0.1, 0.3, 0.5, 0.8, 1, 2vol%) of  $\text{Sm}_2\text{O}_3$

rare-earth oxide additive on the densification, microstructure, and bending strength of  $\text{Al}_2\text{O}_3$  ceramics were investigated to reveal the efficiency of the  $\text{SmAlO}_3$  secondary phase on these properties. The morphology of the formed phase and the mechanical properties (elastic modulus and strength) were investigated depending on the additive ratio and the sintering temperature.

## 2. Materials and Methods

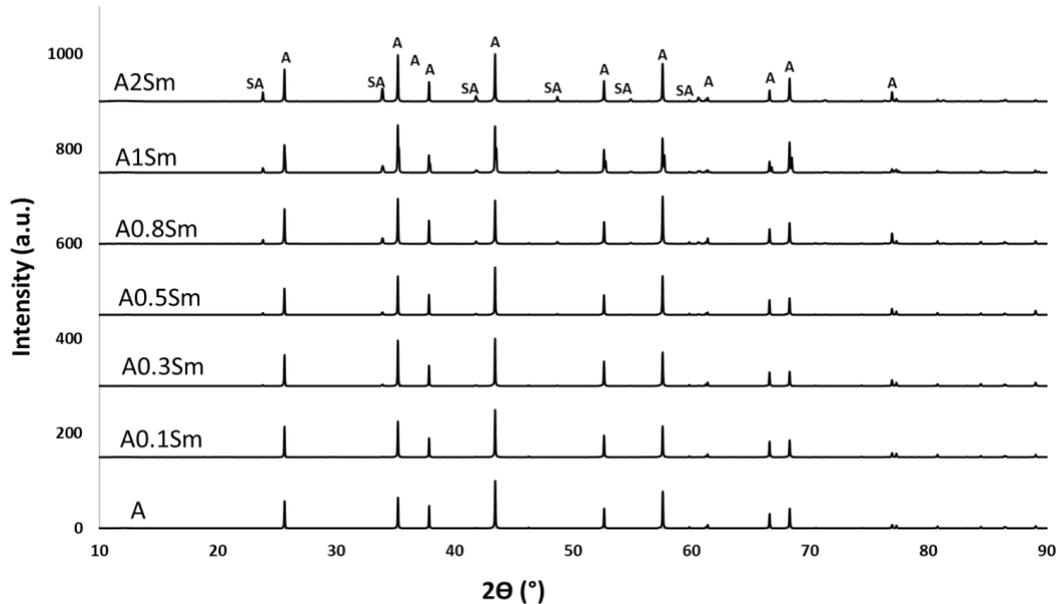
To produce  $\text{Al}_2\text{O}_3$ - $\text{Sm}_2\text{O}_3$  ceramics, a high purity  $\alpha$ - $\text{Al}_2\text{O}_3$  powder (Alfa Aesar, 99.95% purity, 0.25-0.45  $\mu\text{m}$ ),  $\text{Sm}_2\text{O}_3$  powder (99.95%, 50 nm, Nanografi, Turkey), polyacrylic acid (MSE Tech Co. Ltd., Turkey) as a dispersant, polyvinyl alcohol (Sigma Aldrich) as polymer binder, and glycerol as plasticizer (Sigma Aldrich) were used as starting powders and chemicals. High-purity powders were used to examine the effect of the  $\text{Sm}_2\text{O}_3$  additive on densification. Different amounts of  $\text{Sm}_2\text{O}_3$  powder (equivalent to 0, 0.1, 0.3, 0.5, 0.8, 1, 2 vol%) and  $\text{Al}_2\text{O}_3$  powder were mixed in an HDPE bottle with distilled water and dispersant (0.5wt%) by ball milling, separately. After 1.5wt% binder + 0.5wt% plasticizer aqueous solution addition to the ball milled powder mixture, the slurry was dried on a hot plate and in a drying oven at 70°C. The dried powder mixture was crushed and sieved through a 170 mesh screen. Then, the powder mixtures were pressed under a uniaxial applied pressure of 40 MPa in a 35 mm diameter cylinder mold. Subsequently, the green bodies taken into the vacuum in a latex material were exposed to cold isostatic pressing under 200 MPa pressure. The prepared specimens were pressureless sintered at 1550°C and 1600°C in air separately with a heating rate of 5°/min, while the binder burn-out process was carried out at 600°C with a heating rate of 2°/min in an air atmosphere. The specimens were also grinded using a single-sided lapping machine with SiC abrasive powder to make equal thicknesses of the pellets with  $\pm 0.05$  mm tolerance for compatibility with the mechanical tests.

The phases present in the sintered specimens were characterized by the X-ray diffraction method (Bruker® D8 Advance) for  $2\theta$  between 10° and 90°. The bulk densities were measured by direct measurement of weight and macroscopic dimensions precisely with a micrometer and a caliper. The theoretical densities of the ceramics were determined by the rule of mixtures (Hossen et al., 2014). After grinding and polishing, the specimens were thermally etched at 100°C below the sintering temperatures for 90 min for microstructural exploration. The grain morphology and average  $\text{Al}_2\text{O}_3$  grain size were characterized by using scanning electron microscopy (TESCAN Mira3 XMU, Czechia) equipped with energy-dispersive X-ray spectroscopy (EDS) to investigate the chemistry of the phases. The grain size measurements were performed with the linear intercept method where more than 100 intercepts were counted for each  $\text{Al}_2\text{O}_3$ - $\text{Sm}_2\text{O}_3$  ceramic with different  $\text{Sm}_2\text{O}_3$  content.

The elastic modulus of the polished specimens was measured via an impulse excitation technique (GrindoSonic® Mk5) appropriate for ASTM E 1876 standard to disc-formed samples. The bending strength test of the  $\text{Al}_2\text{O}_3\text{-Sm}_2\text{O}_3$  ceramics was carried out using the equibiaxial flexural strength test procedure consistent with ASTM C 1499 at a loading rate of 0.3 mm/min with 23.9 mm support ring and 9.8 mm load ring test configuration for fifteen specimens at ambient temperature using a universal test machine (Instron-5569, USA).

### 3. Findings and Discussion

XRD diffraction patterns of the  $\text{Al}_2\text{O}_3\text{-Sm}_2\text{O}_3$  ceramics prepared with different  $\text{Sm}_2\text{O}_3$  volume ratios are given in Figure 1. A single graph is given as the XRD diffraction patterns are almost the same for the samples sintered at 1550°C and 1600°C. In the graph, “A” indicates  $\text{Al}_2\text{O}_3$  peaks while “SA” refers to  $\text{SmAlO}_3$  peaks. The presence of  $\text{Al}_2\text{O}_3$  ( $\alpha$ -type, hexagonal corundum) and  $\text{SmAlO}_3$  phase (orthorhombic) formed with the reaction of  $\text{Al}_2\text{O}_3$  and  $\text{Sm}_2\text{O}_3$  were detected. No other extra phase was observed, and  $\text{SmAlO}_3$  peaks became evident, especially after 0.5vol%  $\text{Sm}_2\text{O}_3$  ratio. When the amount of  $\text{Sm}_2\text{O}_3$  increased, the peaks of the  $\text{SmAlO}_3$  phase gradually became apparent.



**Figure 1.** The XRD diffraction patterns of the  $\text{Al}_2\text{O}_3\text{-Sm}_2\text{O}_3$  ceramics prepared with different  $\text{Sm}_2\text{O}_3$  volume ratios.

The relative density,  $\text{Al}_2\text{O}_3$  grain size values, and mechanical properties are shown depending on the  $\text{Sm}_2\text{O}_3$  amount and the sintering temperature in Table 1. The theoretical density values were estimated by the rule of mixtures, and the relative densities were calculated from the ratio of the bulk densities to the theoretical densities. Almost similar relative density values were obtained in the

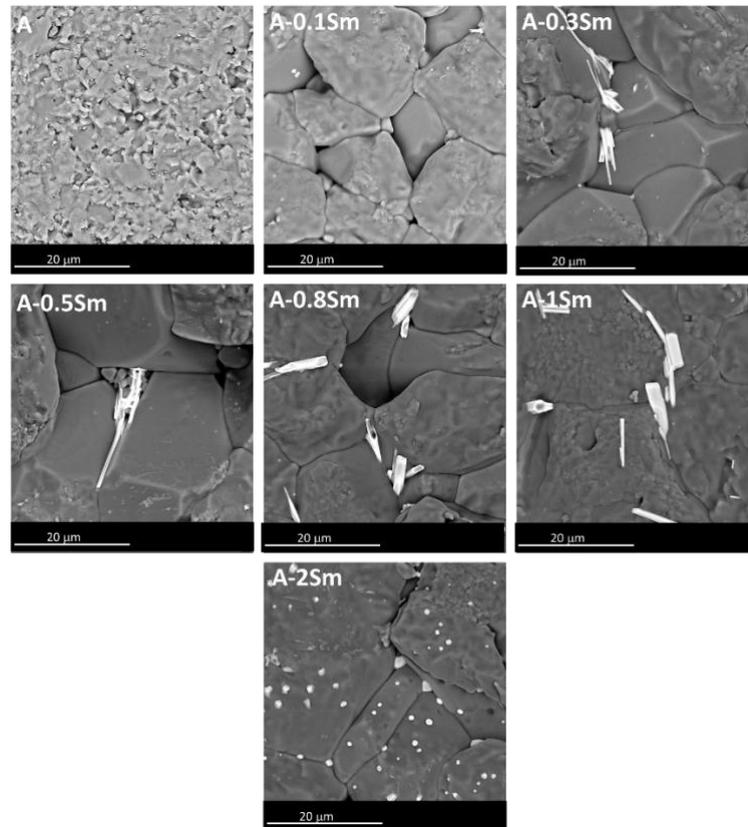
$\text{Al}_2\text{O}_3$ - $\text{Sm}_2\text{O}_3$  ceramics for all the  $\text{Sm}_2\text{O}_3$  ratios at both sintering temperatures. However, the pure  $\text{Al}_2\text{O}_3$  samples sintered at  $1550^\circ\text{C}$  exhibited higher densification than the samples sintered at  $1600^\circ\text{C}$ . In general, it was observed that the addition of  $\text{Sm}_2\text{O}_3$  for sintering at  $1550^\circ\text{C}$  did not affect the densification of  $\text{Al}_2\text{O}_3$  considerably. For  $1600^\circ\text{C}$ , it can be thought that there is a slight increase in the relative density values after the addition of the  $\text{Sm}_2\text{O}_3$  additive. Shi et al. stated in their study that after  $\text{Sm}_2\text{O}_3$  is added, it dissolves into  $\text{Al}_2\text{O}_3$  by taking up the site originally occupied by  $\text{Al}^{3+}$  cations with  $\text{Sm}^{3+}$  cations, increasing the concentration of aluminum vacancies. A higher concentration of aluminum vacancies encourages mass transformation and pores to be eliminated, resulting in significant densification (Shi et al., 2020). This was not determined for this study. It would be useful to view the SEM images of the thermally etched samples in both Figure 2 and Figure 3 to understand the discrepancy. In these figures, “A” denotes the pure alumina, and “Sm” states the samples containing  $\text{Sm}_2\text{O}_3$ , while the number in front of the letter Sm denotes the amount of  $\text{Sm}_2\text{O}_3$  added. Dark gray areas belong to  $\text{Al}_2\text{O}_3$  grains in the SEM figures.

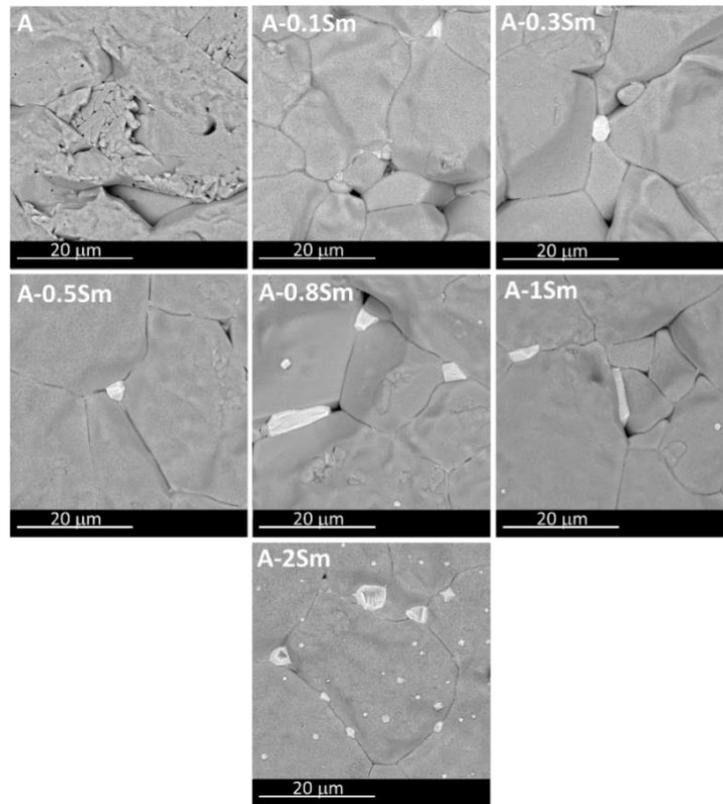
The  $\text{Al}_2\text{O}_3$  powder used in this study is a very high-purity powder and does not contain any grain growth inhibitors and sintering aids such as MgO. In various studies using rare earth oxides such as  $\text{Sm}_2\text{O}_3$ , it has been found that the densification effect of these rare earth additives is more effective in the presence of sintering aids up to the sintering temperatures (Rani et al., 2004). When the sintering temperature rises from  $1550^\circ\text{C}$  to  $1600^\circ\text{C}$ , inhomogeneous particle sizes can be seen for the pure  $\text{Al}_2\text{O}_3$  in Figure 3 with both fine grains and too large grains. Abnormal grain growth was seen in the pure  $\text{Al}_2\text{O}_3$  samples sintered at  $1600^\circ\text{C}$ . In addition to abnormal grain growth, the presence of large porosities in these samples was also detected in the SEM images. Based on these results, it was decided that  $1600^\circ\text{C}$  is not a suitable sintering temperature for the powder used and the present production conditions. For sintering at  $1550^\circ\text{C}$ , the pure  $\text{Al}_2\text{O}_3$  grains were more homogeneous, and an average grain size of  $1.8 \mu\text{m}$  was obtained as given in Table 1. With the addition of  $\text{Sm}_2\text{O}_3$ , the grains grew more homogeneously but reached quite large sizes about  $25\text{-}30 \mu\text{m}$  at both sintering temperatures. No correlation could be specified between the sintering temperature, the  $\text{Sm}_2\text{O}_3$  ratios and the grain size. It is thought that this disorder is related to the formation of the secondary  $\text{SmAlO}_3$  phase and its distribution differs from region to region in the specimens.

The presence of the  $\text{SmAlO}_3$  phase in rod-like form can be easily seen in white color in SEM images for 0.3, 0.5, 0.8, and 1 vol%  $\text{Sm}_2\text{O}_3$  content in Figure 2. However, it is seen that this rod-like form becomes spherical in samples containing 2 vol%  $\text{Sm}_2\text{O}_3$  for both sintering temperatures. In the samples sintered at  $1600^\circ\text{C}$ , the rod-like form was obtained mostly for 0.8 and 1 vol%  $\text{Sm}_2\text{O}_3$  ratios. On the contrary, the rod-like form was more pronounced for the samples sintered at  $1550^\circ\text{C}$ . These results showed that the morphology of the  $\text{SmAlO}_3$  phase was related to the  $\text{Sm}_2\text{O}_3$  amount and the sintering temperature.

**Table 1:** Densification, matrix grain size, and mechanical properties of the prepared Al<sub>2</sub>O<sub>3</sub>-Sm<sub>2</sub>O<sub>3</sub> ceramics for different sintering temperatures.

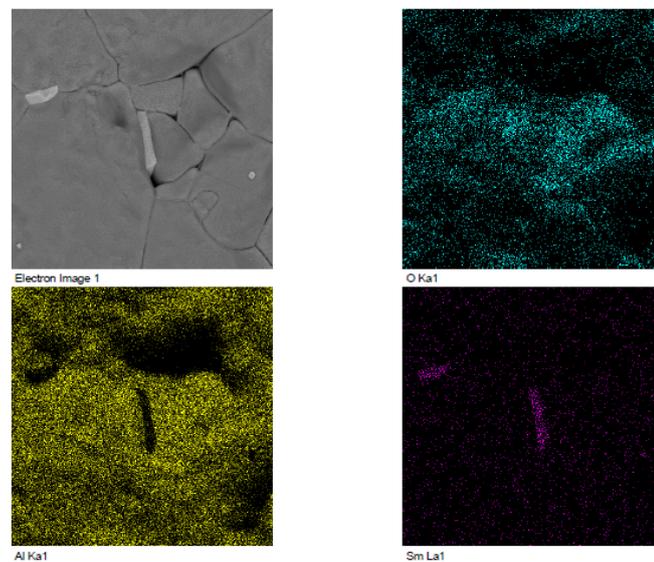
Sm <sub>2</sub> O <sub>3</sub> content (vol%)	Sintering Temperature (°C)	Relative Density (%)	Elastic Modulus (GPa)	Bending Strength (MPa)	Average Al <sub>2</sub> O <sub>3</sub> grain size (mm)
<b>0</b>	<b>1550</b>	<b>98.4 ± 0.3</b>	<b>390 ± 3</b>	<b>279 ± 27</b>	<b>1.8</b>
0.1	1550	98.5 ± 0.2	388 ± 2	205 ± 15	29.5
0.3	1550	98.4 ± 0.4	389 ± 3	198 ± 19	25.7
0.5	1550	98.6 ± 0.3	389 ± 5	193 ± 17	30.2
0.8	1550	98.6 ± 0.3	390 ± 2	189 ± 14	36.5
1	1550	98.4 ± 0.4	385 ± 6	191 ± 12	26.9
2	1550	98.2 ± 0.2	383 ± 7	177 ± 13	27.4
<b>0</b>	<b>1600</b>	<b>97.8 ± 0.5</b>	<b>381 ± 5</b>	<b>188 ± 15</b>	<b>8.3</b>
0.1	1600	98.1 ± 0.5	386 ± 5	195 ± 12	21.1
0.3	1600	98.3 ± 0.4	385 ± 5	192 ± 8	26.6
0.5	1600	98.3 ± 0.4	386 ± 3	197 ± 14	30.2
0.8	1600	98.4 ± 0.3	388 ± 3	190 ± 11	29.4
1	1600	98.2 ± 0.3	384 ± 3	185 ± 13	33.4
2	1600	98.2 ± 0.4	383 ± 5	175 ± 15	32.5

**Figure 2.** SEM images of the thermally etched Al<sub>2</sub>O<sub>3</sub>-Sm<sub>2</sub>O<sub>3</sub> ceramics sintered at 1550°C.



**Figure 3.** SEM images of the thermally etched  $\text{Al}_2\text{O}_3\text{-Sm}_2\text{O}_3$  ceramics sintered at  $1600^\circ\text{C}$ .

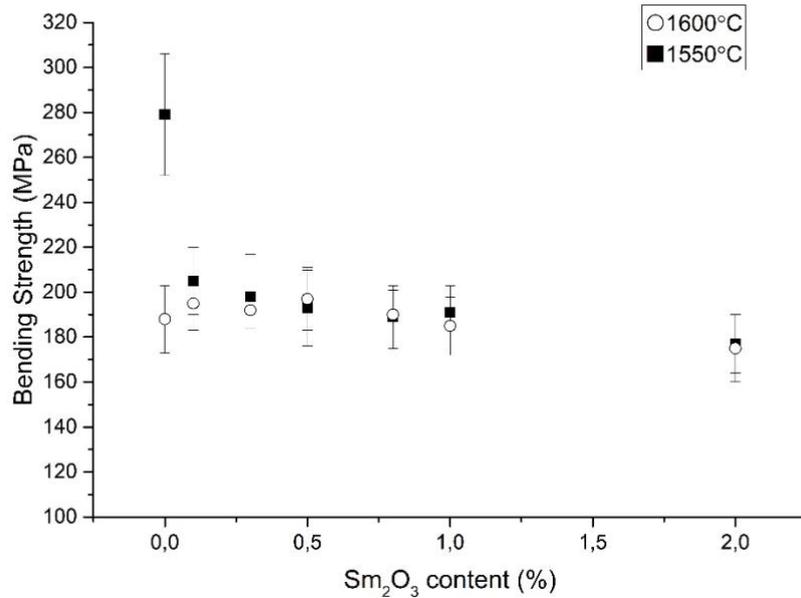
The SEM-EDX mapping results of the thermally etched  $\text{Al}_2\text{O}_3\text{-1vol\%Sm}_2\text{O}_3$  containing ceramic are presented in Figure 4. The turquoise areas belong to O elements, while the yellow and purple regions indicate Al and Sm elements, respectively. From the SEM-EDX results, it was more clearly determined that the phase in rod-like form belongs to  $\text{SmAlO}_3$  particles that were mostly positioned at the triple junctions and grain boundaries.



**Figure 4.** SEM-EDX mapping images of the thermally etched  $\text{Al}_2\text{O}_3\text{-1vol\%Sm}_2\text{O}_3$  containing ceramic.

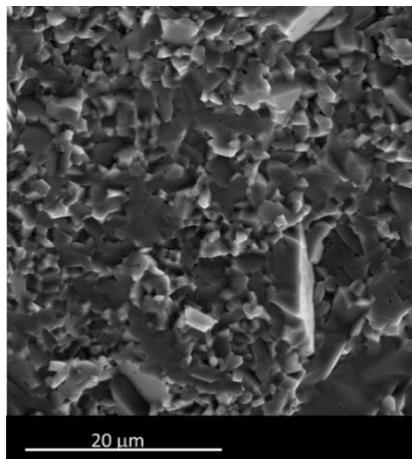
When the produced ceramics are examined in terms of mechanical properties, it is seen that the elastic modulus results were nearly the same for all the produced ceramics. Since the relative density results were similar, it can be thought that the elastic modulus values were also similar as the densification ratio affects the elastic modulus of a ceramic material. It is stated in the literature that the elastic modulus of a material is greatly affected by porosity (Feng et al., 2019). Therefore, the lowest elastic modulus was obtained for the pure  $\text{Al}_2\text{O}_3$  sintered at  $1600^\circ\text{C}$  with a value of 381 GPa.

The bending strength values of  $\text{Al}_2\text{O}_3\text{-Sm}_2\text{O}_3$  ceramics as a function of  $\text{Sm}_2\text{O}_3$  volume content are given in Figure 5 and Table 1 for both sintering temperatures. Although the strength values are generally close to each other, a strength increase of about 5% was obtained for 0.5 vol%  $\text{Sm}_2\text{O}_3$  containing ceramics sintered at  $1600^\circ\text{C}$  due to the higher densification compared to the pure  $\text{Al}_2\text{O}_3$  sintered at  $1600^\circ\text{C}$ . For the 2vol%  $\text{Sm}_2\text{O}_3$  ratio, a decrease of ~7% was observed in the strength for  $1600^\circ\text{C}$ . Although the strength showed further decline above the ratio of 0.1 vol%  $\text{Sm}_2\text{O}_3$ , the strength values were nearly similar up to 2vol%  $\text{Sm}_2\text{O}_3$  content for the other ratios (0.3, 0.5, 0.8, 1 vol%) for sintering at  $1550^\circ\text{C}$ . It is thought that the more significant decrease in strength for the 2vol%  $\text{Sm}_2\text{O}_3$  ratio at both sintering temperatures is related to the morphology of the  $\text{SmAlO}_3$  phase becoming completely spherical from the rod-like form at this ratio by including the  $\text{SmAlO}_3$  phase embedded in the  $\text{Al}_2\text{O}_3$  grains. Besides, the reason for the decrease in bending strength with the addition of  $\text{Sm}_2\text{O}_3$  is the presence of quite large  $\text{Al}_2\text{O}_3$  grains for sintering at  $1550^\circ\text{C}$ . It is stated that the bending strength of a ceramic material relates strongly to its grain size and flaw size. Larger grains introduce greater flaws, which results in lower strength. (Tuan et al., 2008). Therefore, larger  $\text{Al}_2\text{O}_3$  grain sizes resulted in lower strength values compared to the pure  $\text{Al}_2\text{O}_3$  for  $\text{Sm}_2\text{O}_3$  containing samples with similar densification for sintering at  $1550^\circ\text{C}$ . Since the pure  $\text{Al}_2\text{O}_3$  sintered at  $1600^\circ\text{C}$  has abnormally grown grains, there was no significant difference in strength between the samples containing  $\text{Sm}_2\text{O}_3$  sintered at this temperature. On the other hand, a direct relationship between the grain size and the bending strength change could not be determined for all the samples containing the  $\text{Sm}_2\text{O}_3$  additive since the grain size values varied. For example, for the samples sintered at  $1600^\circ\text{C}$ , 197 MPa strength was measured in the sample containing 0.5vol%  $\text{Sm}_2\text{O}_3$  with a grain size of approximately 30  $\mu\text{m}$ , while almost the same 195 MPa strength was obtained in the sample containing 0.1 vol%  $\text{Sm}_2\text{O}_3$  with a grain size of 21  $\mu\text{m}$ .



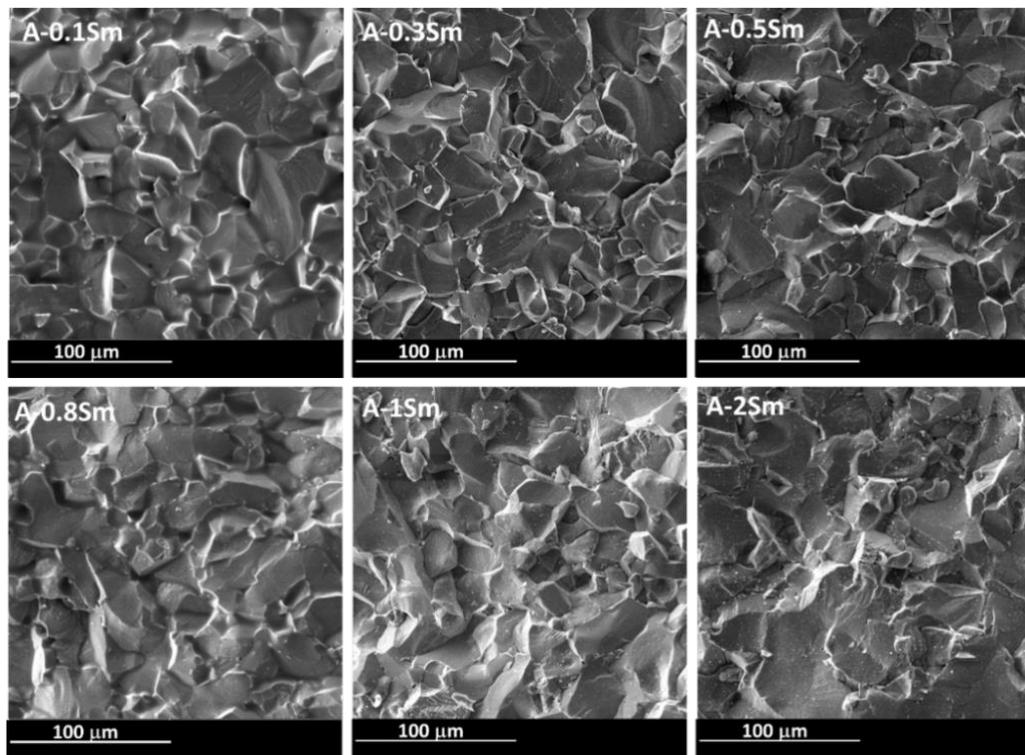
**Figure 5.** The bending strength of Al<sub>2</sub>O<sub>3</sub>-Sm<sub>2</sub>O<sub>3</sub> ceramics as a function of Sm<sub>2</sub>O<sub>3</sub> volume content for the sintering temperature.

In order to interpret the fracture strength behavior depending on the microstructure, it is also necessary to examine the fracture surface SEM images. Figure 6 shows the fracture surface of the pure Al<sub>2</sub>O<sub>3</sub> sintered at 1550°C. Although the fracture mode was mostly intergranular for the small grains, it is seen that the coarser grains were fractured by cleavage. It is stated that when the grain size of a ceramic material is enhanced, the fracture mode alters from intergranular to cleavage (transgranular). As ceramics often have fine grain sizes, their grain boundaries have a larger surface area and more energy than the grains themselves, which causes intergranular fracture most of the time. However, with an increase in grain size, grain boundary area decreases thus changing the fracture mode to cleavage (Kambale et al., 2019).

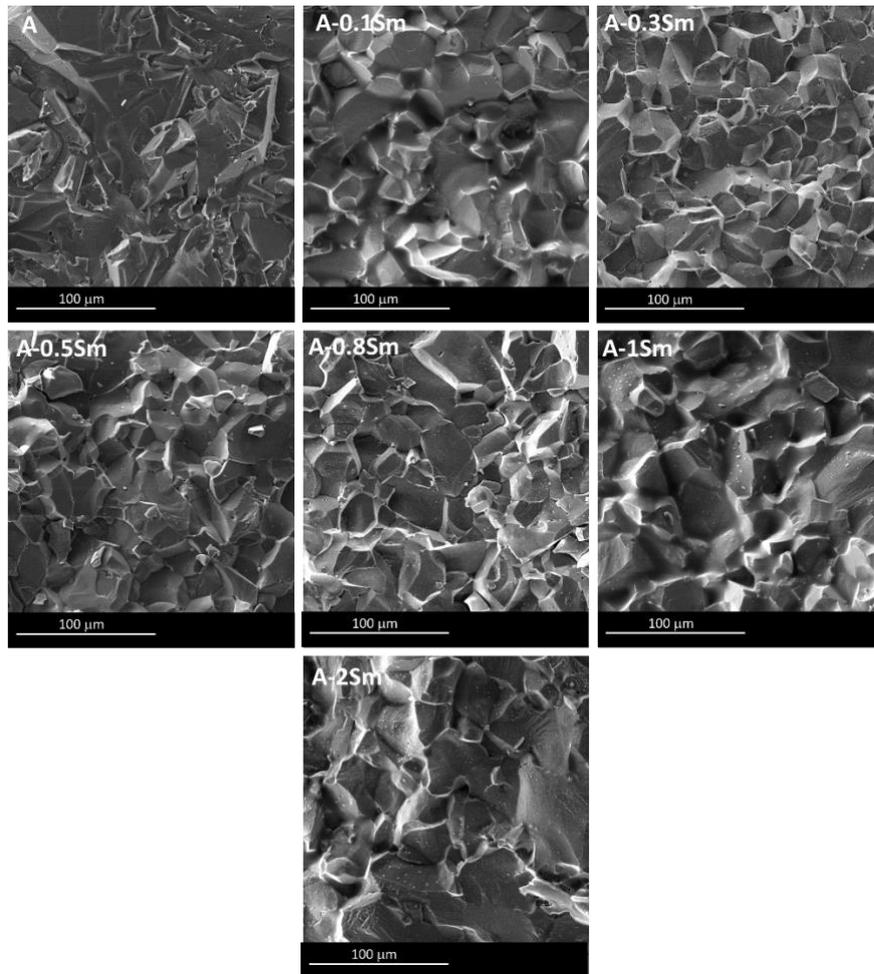


**Figure 6.** SEM image of the fracture surface of the Al<sub>2</sub>O<sub>3</sub> sintered at 1550°C.

SEM images of the fracture surfaces of the  $\text{Al}_2\text{O}_3\text{-Sm}_2\text{O}_3$  ceramics for both sintering temperatures are also given in Figure 7 and Figure 8, separately. For sintering at  $1550^\circ\text{C}$ , the  $\text{Al}_2\text{O}_3\text{-Sm}_2\text{O}_3$  ceramics for all the  $\text{Sm}_2\text{O}_3$  additive ratios showed almost completely transgranular fracture mode differently from the pure  $\text{Al}_2\text{O}_3$  sintered at the same temperature. This can clearly be attributed to the much larger size of the  $\text{Al}_2\text{O}_3$  grains in the  $\text{Al}_2\text{O}_3\text{-Sm}_2\text{O}_3$  ceramics compared to the pure  $\text{Al}_2\text{O}_3$ . The possible effect of the formed  $\text{SmAlO}_3$  phase on the fracture behavior could not be identified due to the effect of large grains. A similar fracture behavior as transgranular was also valid for the  $\text{Al}_2\text{O}_3\text{-Sm}_2\text{O}_3$  ceramics sintered at  $1600^\circ\text{C}$ . In addition, the difference in the fracture mode for the pure  $\text{Al}_2\text{O}_3$  with abnormal growth grains for sintering at  $1600^\circ\text{C}$  is seen compared to Figure 6 for sintering at  $1550^\circ\text{C}$ .



**Figure 7.** SEM images of the fracture surfaces of the  $\text{Al}_2\text{O}_3\text{-Sm}_2\text{O}_3$  ceramics sintered at  $1550^\circ\text{C}$ .



**Figure 8.** SEM images of the fracture surfaces of the  $\text{Al}_2\text{O}_3\text{-Sm}_2\text{O}_3$  ceramics sintered at  $1600^\circ\text{C}$ .

#### 4. Conclusions

In the present investigation, the effects of different amounts (0, 0.1, 0.3, 0.5, 0.8, 1, 2vol%) of  $\text{Sm}_2\text{O}_3$  additive on the densification, microstructure, and mechanical properties (elastic modulus and bending strength) of  $\text{Al}_2\text{O}_3$  ceramics were studied for different sintering temperatures. The fabrication procedure involved ball milling of nano-sized  $\text{Sm}_2\text{O}_3$  powder with the  $\text{Al}_2\text{O}_3$  powder in proper volume ratios and followed by pressureless sintering at  $1550^\circ\text{C}$  and  $1600^\circ\text{C}$  for two hours in the air separately after uniaxially dry and cold isostatic pressing.  $\text{Sm}_2\text{O}_3$  reacted with the  $\text{Al}_2\text{O}_3$  by forming the  $\text{SmAlO}_3$  phase at the sintering temperatures and the existence of the  $\text{SmAlO}_3$  phase was confirmed by the XRD analysis. In the samples sintered at  $1600^\circ\text{C}$ , the rod-like form of the  $\text{SmAlO}_3$  phase was obtained mostly for 0.8 and 1vol%  $\text{Sm}_2\text{O}_3$  ratios while the rod-like form of the phase was more evident for the samples sintered at  $1550^\circ\text{C}$ . Similar densification behavior was obtained in the  $\text{Al}_2\text{O}_3\text{-Sm}_2\text{O}_3$  ceramics for all  $\text{Sm}_2\text{O}_3$  ratios at both sintering temperatures. Nevertheless, the pure  $\text{Al}_2\text{O}_3$  sintered at  $1550^\circ\text{C}$  exhibited higher densification than the samples sintered at  $1600^\circ\text{C}$ . Since the relative density results were similar, the elastic modulus values were also parallel. Though the

bending strength values were in general close to each other, a strength increase of about 5% was obtained for 0.5 vol% Sm<sub>2</sub>O<sub>3</sub> containing ceramics sintered at 1600°C caused by the higher densification compared to the pure Al<sub>2</sub>O<sub>3</sub>. The strength values showed a further drop above the ratio of 0.1 vol% Sm<sub>2</sub>O<sub>3</sub>, but they were nearly similar up to 2vol% Sm<sub>2</sub>O<sub>3</sub> content for the other ratios (0.3, 0.5, 0.8, 1vol%) for sintering at 1550°C. For both sintering temperatures, the Al<sub>2</sub>O<sub>3</sub>-Sm<sub>2</sub>O<sub>3</sub> ceramics in all Sm<sub>2</sub>O<sub>3</sub> additive ratios showed transgranular fracture mode. Based on the results obtained, it was observed that the addition of Sm<sub>2</sub>O<sub>3</sub> to the relevant ratios did not cause a remarkable increase in the mechanical properties of the Al<sub>2</sub>O<sub>3</sub> ceramic but changed the microstructure.

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### **Authors' Contributions**

ST: Experimental, Investigation, Visualization. BKY: Conceptualization, Investigation, Experimental, Writing - review and editing. EI: Experimental, Investigation. YKT: Conceptualization, Writing - review and editing.

### **Statement of Conflicts of Interest**

There is no conflict of interest between the authors.

### **Statement of Research and Publication Ethics**

The author declares that this study complies with Research and Publication Ethics.

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