

# Exergetic and Economic Assessment of Distillation Hybrid Configurations for Bioethanol Refining

B. Suleiman\*, A.S Olawale<sup>+</sup> and S.M. Waziri<sup>+</sup>

\*Department of Chemical Engineering, Federal University of Technology Minna, Nigeria.

<sup>+</sup>Department of Chemical Engineering, Ahmadu Bello University Zaria, Nigeria

\*Corresponding Author E-mail: bilyaminusuleiman@yahoo.com

## Abstract

Thermo-economics analysis was used to identify the most economic distillation hybrid configuration to dehydrate bioethanol mash (12 wt%) to fuel grade (99.5 wt%) based on economic objective of minimization of operating cost in this work. Three different hybrids of THIDC with azeotropic and, extractive distillation units were assessed using similar feed and product specifications of 1200 kmol/h (12 wt% ethanol) and 55 kmol/h (99.5 wt% ethanol) respectively. The six hybrid configurations were simulated using Aspen Plus®. The hybrid of THIDC with conventional extractive distillation (THEX1) was shown to have the lowest irreversibility rate (lost work) and highest exergetic efficiency followed by the hybrid containing thermally extractive sequence (THEX3). The latter also has the lowest energy consumption. However, economic evaluation showed that thermally coupled extractive distillation hybrid (with THIDC) is the most attractive hybrid configuration dehydrating bioethanol to fuel grade at commercial scale with the highest return on investment (ROI) and the least annual product cost. This indicates its economic attractiveness when compared with the other hybrids considered in this work. The trade-off existing between economic and exergy efficiency favors the selection of THEX3 as the preferred choice for bioethanol refining among all the six hybrids investigated.

**Keywords:** Totally heat integrated distillation column; lost work; azeotropic distillation; extractive distillation; exergy efficiency; product cost.

## 1. Introduction

Anhydrous ethanol production has become one of the most important issues all over the world. This is due to the great efforts directed to the use of biofuels and reduction in pollution and environmental effects of fossil fuels.

The process of anhydrous ethanol production comprises three main steps: fermentation, distillation and dehydration [1]. Bioethanol can be produced from sugar, starch or cellulosic feed stock by fermentation. The major problem associated with refining of resulting ethanol to fuel grade is the energy consumption cost. The 10-20 wt% ethanol obtained after fermentation [2,3] could only be distilled to maximum of 95.6 wt% due to the formation of azeotrope; thus the removal of part or all of the remaining 4.4 wt% water to obtain fuel grade is usually carried out using one of the following separation techniques: azeotropic distillation, extractive distillation, pervaporation and pressure swing molecular sieves adsorption processes [4].

The use of hybrid separation systems provide means of achieving cheaper, easier and enhanced separation by linking distillation unit to any of the dehydration processes mentioned above.

Exergetic and economic analysis provides useful information for identification and quantitative measure of the thermodynamic imperfections in processes (as a result of production of entropy) as well as cost implication of the operations. The result of thermodynamic (exergy) analysis may be in line with those of economic analyses when the thermodynamic cost optimum takes precedence over maximum thermodynamic efficiency in process specification [5]. The ultimate aim of the analysis is to

obtain an energy efficient process possessing minimum capital and operating cost.

Many works in this field focused on reduction in energy consumption and efficiency improvements of either a standalone unit or the hybrid unit (i.e. distillation and dehydration units). In the area of energy requirement several researchers worked to reduce energy consumption of either distillation stand alone or hybrid units containing azeotropic, extractive or pervaporation unit [1,6,7,8,2]. The hybrid configurations of distillation with three extractive sequences were compared based on energy requirement and second law efficiency on the basis of similar feed and product specifications [9]. Several attempts were made to study and investigate bioethanol refining using molecular sieves and to investigate the effects of several operating parameters for process improvements [10,11,12,13]. Bioethanol refining was also studied using hybrid distillation-pervaporation process and the serially connected module was chosen as the best [14]. In addition, distillation-pervaporation hybrid was studied using combined Aspen-Plus® and Excel visual basic for application which provided not only design tool but optimization procedure [15].

Furthermore, works in energy and economic analysis of bioethanol refining processes were carried out for the purpose of techno-economic comparison of energy usage between hybrids. Among such is the techno-economic comparison of energy usage between hybrids of azeotropic distillation and pervaporation with conventional distillation column [16]. It was found that the hybrid incorporating pervaporation consumes 52.4 % less energy than that

containing azeotropic distillation. Analysis was carried out to establish the most energy efficient among conventional, Petlyuk and thermally coupled extractive distillation sequences for bioethanol purification to 99.5 wt% [17]. Thermally extractive sequence recorded about 30 % reduction in energy consumption. However, energy saving and capital cost evaluation in distillation column sequences with divided wall was carried out and showed that divided wall column gave significant energy reduction with less than 30 % capital cost savings [18].

In all of these previous works the choice of the distillation column type that formed the hybrid with the dehydrating unit appeared to have been made arbitrarily.

However, totally heat integrated distillation column (THIDC) was identified as the best of the distillation column types for bioethanol separation [19]. The other four distillation column types studied were the simple conventional column, distillation column having intermediate heat exchanger with and without heat pump and secondary reflux and vaporization distillation column (SRVP). Comparison of various distillation based hybrid configurations that incorporate THIDC in terms of exergetic and economic performance has not been carried out for bioethanol refining. Therefore in this work different hybrid configurations of THIDC with azeotropic or extractive distillation unit were assessed from exergetic-economic view point. Similar feed and product specifications were used for the six distillation hybrid configurations investigated.

## 2. Methodology

Three azeotropic and extractive distillation configurations were selected from the works reported in the literature [7,16,9,2]. For the two dehydrating distillation processes, the selected configurations were the conventional column and two other integrated configurations which perform relatively very well in term of product purity, throughputs and energy consumption. They were individually connected to THIDC (using feed at 102 °C) to obtain the hybrid configurations for dehydrating the water-ethanol mixture. These six hybrid configurations were subjected to simulation using Aspen Plus® to generate data for exergy and economic analysis. The six configurations were then ranked in order to select the best configuration based on economic consideration.

### 2.1 Simulation of THIDC-Azeotropic Distillation

#### Hybrid Configurations

Three different types of azeotropic distillation columns (as shown in Figures 1-3) were used to form hybrid with THIDC which was identified as the best [19]. The configurations were those whose recycle stream enters into: (a) azeotropic column with organic phase (Figure 1), (b) the azeotropic column along with fresh feed (Figure 2), (c) the decanter (Figure 3). These three azeotropic units have been identified as the best [2].

In all cases a *Radfrac* column type was selected from Aspen-Plus® window in developing each flow sheet. The simulated THIDC column was replicated by exporting the developed flow sheet to a new Aspen-Plus® simulation environment. The third component - an entrainer (cyclohexane) - was added in the components list specified for the system. Though, benzene is more favourable

economically and energy-wise it is not used anymore because of its environmental consequences [2]. The rectification column (THIDC) distillate (67.9 kmol/h) was connected as the feed stream into each of the three azeotropic distillation column types either as separate feed or combined feed with the recycle stream containing 99.5 wt% ethanol. See Figure 1 as an example.

Each of the azeotropic section consists of dehydration and azeotropic column having internal decanter with the exception of the configuration whose recycle stream was connected to an external decanter (Figure 3). The azeotropic distillate products were cooled in a condenser and split into aqueous and organic phase in the decanter. In all simulations, a feed flow rate of 1200 kmol/h (12 wt% ethanol) into THIDC and azeotropic column bottoms products of 52 kmol/h (99.5 wt% ethanol) was maintained. In addition, the tray sizing, required block and stream specifications to the azeotropic section were provided and the three hybrid configurations were rigorously simulated. The simulation result obtained was transferred to Microsoft excel® (Microsoft office suite) for use in subsequent analysis.

### 2.2 Simulation of THIDC-Extractive Distillation

#### Hybrid Configurations

Three different configurations of extractive distillation were used to form hybrid with totally heat integrated distillation column (THIDC). They were the hybrids containing: (a) conventional extractive distillation (Figure 4), (b) Petlyuk column and (c) thermally coupled column (Figure 6). The THIDC unit specifications were kept fixed with little interaction as the case may be with its downstream.

The hybrid units were simulated as follows: The simulated THIDC column was replicated in a new Aspen Plus® simulation environment and a third component entrainer (ethylene-glycol) was added in the components list specified for the system. The choice of this entrainer was based on its low cost, good capacity and selectivity [21]. THIDC rectification distillate (67.9 kmol/h containing 95.5 wt% ethanol) was connected as the feed stream to extractive columns of configurations 1 and 3 (Figure 1 and 3). For configuration 2 (Figure 2), it was entering as the feed into Petlyuk column. The entrainer stream was recovered as bottom product of the extractive columns of configuration 2 and 3 but recovered as bottom products of the recovery column of configuration 1.

*Radfrac* column type was used for the simulation and extractive column distillate products were maintained as ethanol (99.5 wt %). The destination stage and flow rate of intermediate streams withdrawn to another column were specified and adjusted throughout the simulation. In addition, the flow sheets were completed and all required streams and blocks variables were specified while column tray sizing and report option was made in a similar way to that of the THIDC column. In all simulations, feed flow rate of 1200 kmol/h (12 wt% ethanol) into THIDC and extractive column distillate products of 52 kmol/h containing 99.5 wt% ethanol were maintained. The developed flow sheets were rigorously simulated until convergence was achieved.

In the entire hybrid configurations (Figures 1-6), similar product and feed specifications were used for the purpose

of comparing the configurations on the same basis. This allows selection of variables of the column and other equipment to achieve the target specification. The input

parameters into the Aspen Plus® for all the configurations are shown in Table 2.

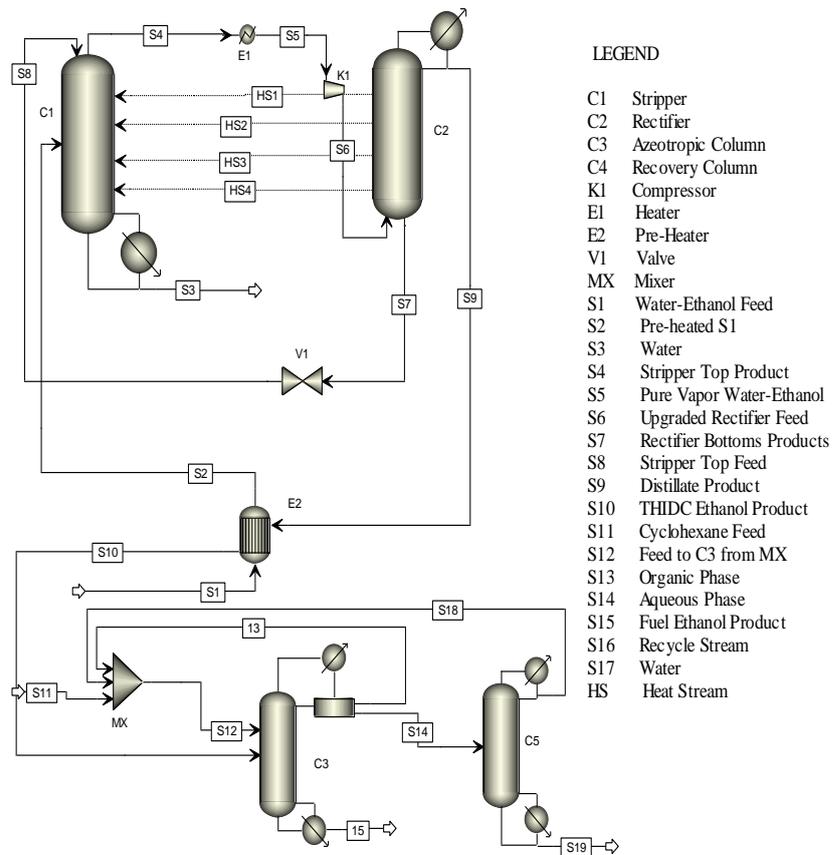


Figure 1. TH1DC-Azeotropic Distillation Hybrid with Recycle Stream Mixing and Entering with Organic Phase (THAZ1)

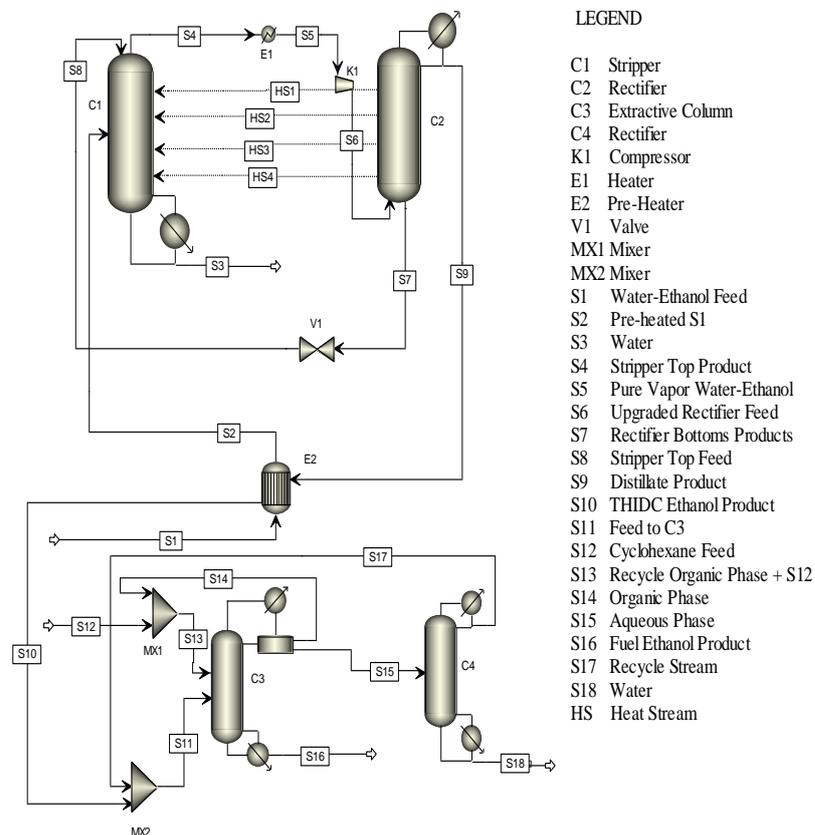


Figure 2. TH1DC-Azeotropic Distillation Hybrid with Recycle Stream Mixing and Entering with Feed (THAZ2)

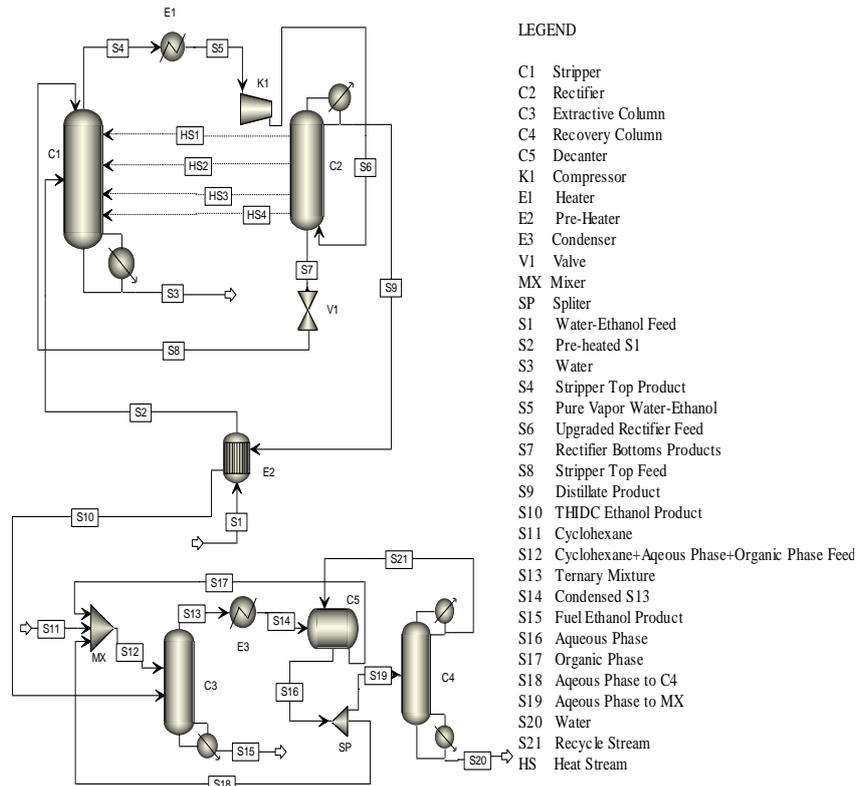


Figure 3. THDC-Azeotropic Distillation Hybrid with recycle Stream Entering into Decanter

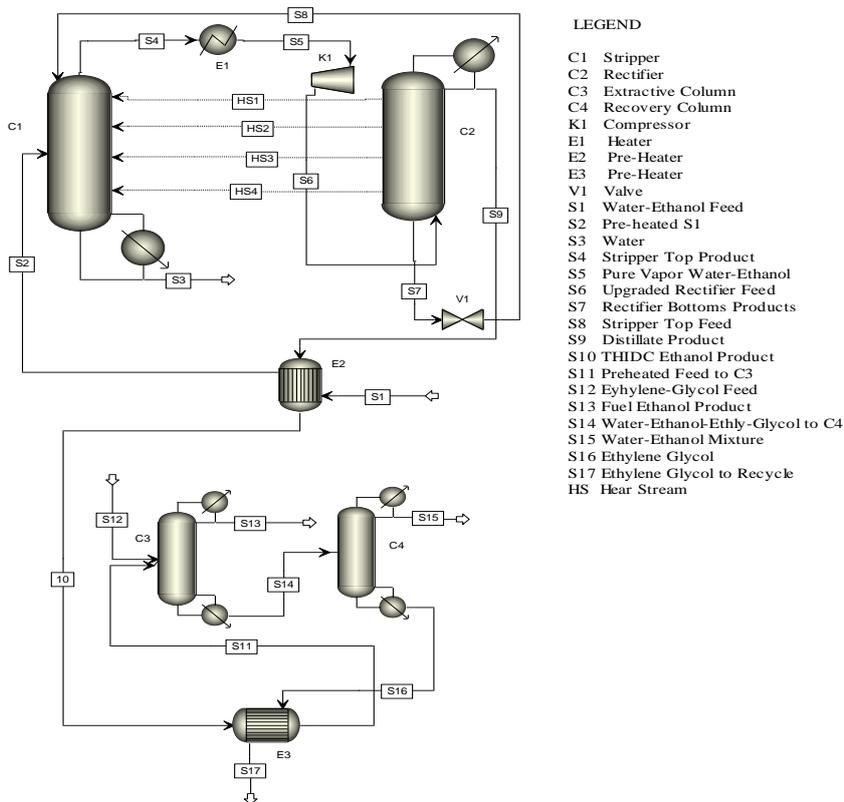
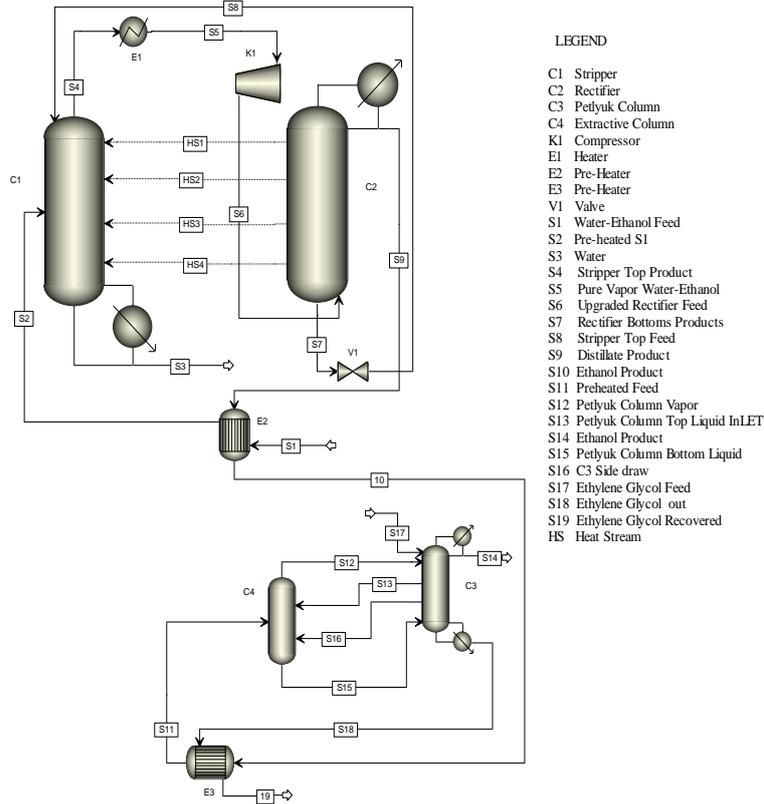
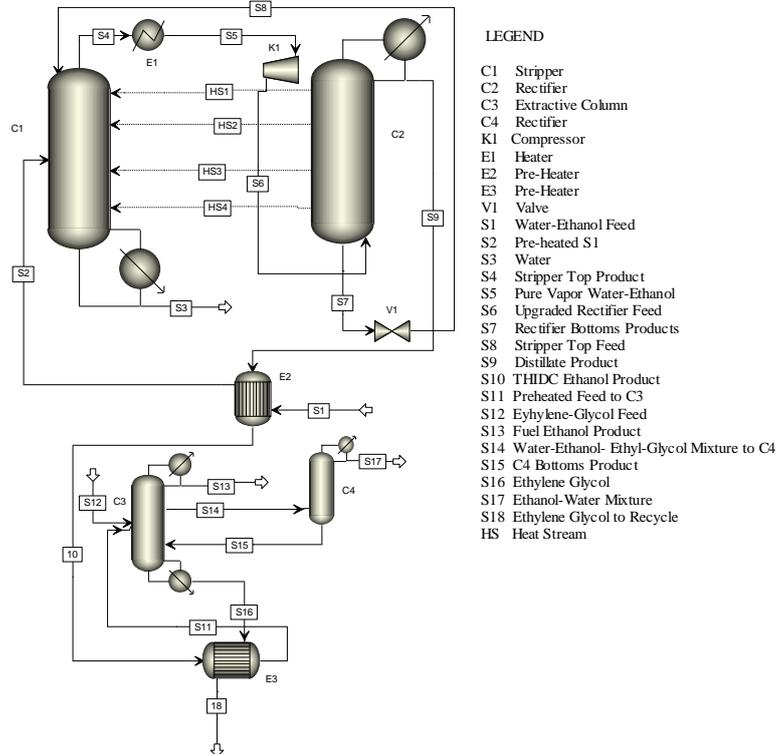


Figure 4. THDC-Conventional Extractive Distillation Hybrid (THEX1)



- LEGEND
- C1 Stripper
  - C2 Rectifier
  - C3 Petlyuk Column
  - C4 Extractive Column
  - K1 Compressor
  - E1 Heater
  - E2 Pre-Heater
  - E3 Pre-Heater
  - V1 Valve
  - S1 Water-Ethanol Feed
  - S2 Pre-heated S1
  - S3 Water
  - S4 Stripper Top Product
  - S5 Pure Vapor Water-Ethanol
  - S6 Upgraded Rectifier Feed
  - S7 Rectifier Bottoms Products
  - S8 Stripper Top Feed
  - S9 Distillate Product
  - S10 Ethanol Product
  - S11 Preheated Feed
  - S12 Petlyuk Column Vapor
  - S13 Petlyuk Column Top Liquid InLET
  - S14 Ethanol Product
  - S15 Petlyuk Column Bottom Liquid
  - S16 C3 Side draw
  - S17 Ethylene Glycol Feed
  - S18 Ethylene Glycol out
  - S19 Ethylene Glycol Recovered
  - HS Heat Stream

Figure 5. THIDC-Petlyuk Column Extractive Distillation Hybrid (THEX2)



- LEGEND
- C1 Stripper
  - C2 Rectifier
  - C3 Extractive Column
  - C4 Rectifier
  - K1 Compressor
  - E1 Heater
  - E2 Pre-Heater
  - E3 Pre-Heater
  - V1 Valve
  - S1 Water-Ethanol Feed
  - S2 Pre-heated S1
  - S3 Water
  - S4 Stripper Top Product
  - S5 Pure Vapor Water-Ethanol
  - S6 Upgraded Rectifier Feed
  - S7 Rectifier Bottoms Products
  - S8 Stripper Top Feed
  - S9 Distillate Product
  - S10 THIDC Ethanol Product
  - S11 Preheated Feed to C3
  - S12 Eethylene-Glycol Feed
  - S13 Fuel Ethanol Product
  - S14 Water-Ethanol- Ethyl-Glycol Mixture to C4
  - S15 C4 Bottoms Product
  - S16 Ethylene Glycol
  - S17 Ethanol-Water Mixture
  - S18 Ethylene Glycol to Recycle
  - HS Heat Stream

Figure 6. THIDC-Thermally Coupled Extractive Distillation Hybrid (THEX3)

### 2.3 Energy and Exergy Analysis of Hybrid Configurations

The data obtained from simulation of the hybrid configurations for water-ethanol mixture dehydration were used for energy and exergy analyses with Microsoft Excel spread sheet software. The parameters determined were

streams' physical and chemical exergy, lost work and exergy efficiency.

The physical exergy  $\dot{B}_{phy}$  (kJ/sec) of a stream of matter with flow in enthalpy  $\dot{h}$  (kJ/sec) and entropy  $\dot{s}$  (kJ/K. sec) at T and P, relative to the surroundings (dead state) at  $T_o$  and  $P_o$  was determined using Eq. 1[21].

$$\dot{B}_{phy} = [\dot{h}(T, P) - \dot{h}(T_o, P_o)] - T_o[\dot{s}(T, P) - \dot{s}(T_o, P_o)] \quad (1)$$

The chemical exergy  $\dot{B}_{chem}$  (kJ/sec) of each stream of matter with component mole fraction  $x_i$  ( $y_i$  for vapour) and standard chemical exergy  $\dot{e}_i^o$  (kJ/sec) of each component  $i$  in a stream was evaluated using Eq. 2 [21].

$$\dot{B}_{chem} = \sum x_i \dot{e}_i^o \quad (2)$$

The total stream exergy  $\dot{B}_{total}$  in Eq. 3 was evaluated as the sum of its physical ( $\dot{B}_{phy}$ ) and chemical exergy ( $\dot{B}_{chem}$ ).

$$\dot{B}_{total} = \dot{B}_{phy} + \dot{B}_{chem} \quad (3)$$

The lost work LW, around each piece of equipment or for overall configuration with streams flows  $\dot{m}$  (kg/s), streams exergy B (kJ/kg), work flows  $\dot{W}$  (kJ/s), utility heat duties Q (kJ/s) at actual temperature T (K) and reference state temperature  $T_o$  (K) was evaluated using Eq. 4 [22].

$$LW = \Sigma(\dot{m}B)_{in} - \Sigma(\dot{m}B)_{out} + \Sigma\dot{W}_{in} - \Sigma\dot{W}_{out} + \Sigma \left[ Q \left( 1 - \frac{T_o}{T} \right) \right]_{in} - \Sigma \left[ Q \left( 1 - \frac{T_o}{T} \right) \right]_{out} \quad (4)$$

In addition, exergetic efficiency ( $\eta_{Exergetic}$ ) was evaluated around equipment and for each hybrid configuration. The exergetic efficiency is the ratio of the exergy recovered  $\Sigma \dot{B}_{out}$  (total output exergy) to the total input exergy  $\Sigma \dot{B}_{in}$  as given in Eq. 5 [23].

$$\eta_{Exergetic} = \frac{\Sigma(Exergy\ out)}{\Sigma(Exergy\ in)} \quad (5)$$

## 2.4 Economic Analysis of Hybrid Configurations

Economic analysis of each of the successfully converged six hybrid configurations was carried out in stages. The first stage was the evaluation of all capital cost of each configuration. Secondly all annual utility, material requirement and cost were determined and subsequently the economic evaluation. However, the whole economic analysis was carried out using the approach developed for chemical plant economic evaluation [24]. Therefore, using this approach, fixed capital and working capitals were obtained using percentage of equipment delivery method.

In addition, parameters in Table 1 were used in the evaluation of materials input cost, equipment cost, utility cost and annual total product sales. However, the utility requirement and its cost was obtained from the Aspen Plus by specifying the heating/cooling duty (kJ/kg), inlet and outlet utility temperature as well as purchased price. The duty of a utility with specific heat capacity  $C_p$  (kJ/kg K) and temperature difference  $\Delta T$  between its inlet and outlet was obtained from the energy balance equation as the product  $C_p \times \Delta T$ .

All costs and products value were subjected to profitability analysis using online spread sheet for chemical plant economic evaluation [27]. Equations used to evaluate profitability parameters are contained in plant design and economics for chemical engineers [25] and the spreadsheets for carrying out this analysis are available online [27].

Table 1. Parameters Assumed for Economic Analysis

S/N	Parameter	Value
1	Operating time (days/yr.)	335
2	Equipment cost basis (CEPCI)	2012
3	Desired product ethanol cost (\$/kg)	4.0
4	Steam cost (Medium pressure, \$/kg)	0.00966
5	Cooling water cost (\$/kg)	0.00063
6	Cyclohexane cost (\$/kg)	0.37
7	Ethylene glycol cost (\$/kg)	0.39
8	Input ethanol cost (\$/kg)	0.02

Table2. Aspen Plus Input Parameters for Simulation of the Hybrids

Parameter	Value					
	THEX1	THEX2	THEX3	THAZ1	THAZ2	THAZ3
Fermenters feed flow (kmol/h)	1200	1200	1200	1200	1200	1200
Azeotropic feed flow (kmol/h)	67.9	67.9	67.9	67.9	67.9	67.9
Separating agent flow (kmol/h)	27.3	36.364	19.5	14	30.3	10.34
Extractive column's distillate flow (kmol/h)	52	52	52	-	-	-
Azeotropic column's bottoms flow (kmol/h)	-	-	-	52	52	52
Molar fraction of fuel-ethanol	0.9899	0.9899	0.9899	0.9899	0.9899	0.9899
Temperature of fermenters feed (°C)	102	102	102	102	102	102
Temperature of azeotropic feed (°C)	105.15	105.15	105.15	106.14	106.14	106.14
Temperature of separating agent (°C)	85	85	85	80	80	80
Molar reflux ratio in rectifier column	4.8	4.8	4.8	5	5	5
Molar reflux ratio in extractive column	0.359	0.356	0.455	-	-	-
Molar boil-up ratio in stripper column	0.148	0.149	0.148	0.148	0.148	0.148
Molar boil-up ratio in azeotropic column	-	-	-	8.541	6.973	2.95
Number of theoretical stages ( C1 column)	30	30	30	30	30	30
Number of theoretical stages ( C2 column)	27	27	27	27	27	27
Number of theoretical stages ( C3 column)	40	48	38	70	74	80
Number of theoretical stages ( C4 column)	12	8	8	12	8	9
Fermenters feed stage	15	15	15	15	15	15
Azeotropic feed stage	28	3	27	45	9	65
Pressure in the C1 column (atm)	1.0023	1.0023	1.0023	1.0023	1.0023	1.0023
Pressure in the C2 column (atm)	17	17	17	19	19	19
Pressure in the C3 column (atm)	1.21	1.6	3.01	2.6	2.55	1.12
Pressure in the C4 column (atm )	1.0	1.6	3.0	1.12	0.7	1.1
Pressure in the decanter (atm)	-	-	-	2.6	1.1	1.0
Separating-agent feed stage	3	4	3	7	7	6
Solvent/feed ratio (S/F)	0.402	0.54	0.287	2E-6	0.446	0.1523
Feed stage of aqueous mixture	-	-	-	8	6	4
Compressor power requirement (kW)	917.5	917.5	917.5	958.7	958.7	958.7
Exchanger 1 duty(kW)	25.31	25.31	25.31	23.87	23.87	23.87
Exchanger 2 duty(kW)	63.22	63.22	62.22	60.29	60.29	60.29
Exchanger 3 duty(kW)	5.21	4.617	22.19	-	-	-

### 3. Results and Discussion

#### 3.1 Energy and Exergy Analysis

Tables 3-6 show the energy and exergy analysis result for the six hybrid configurations selected in this work. The material streams and utility exergy flows in and out were used to carry out exergy analysis for each of the hybrid unit. The various streams exergy results include the contributions of physical and chemical exergy evaluated using Eq. (1) and (3). The exergy of mixing was included in physical exergy evaluation since the respective streams entropies were used.

Table3. Total Input Energy, Exergy efficiency and Lost Work for all Hybrids

Configuration	Input energy (kW)	Lost work (kW)	Exergy Efficiency (fraction)
THAZ1	37700	13400	0.583
THAZ2	69700	39900	0.383
THAZ3	47900	23000	0.466
THEX1	39200	2644	0.927
THEX2	42202	4010	0.901
THEX3	36666	2730	0.918

Table 3 shows the energy and exergy analysis results for the six different hybrid configurations formed by linking THIDC to azeotropic and extractive distillation

units. The total input energy includes energy associated with material feed as well as thermal and work energy inputs. The THIDC-extractive sequences hybrids are shown to be the most efficient. The least efficient hybrid was THAZ2 with the highest lost work and least exergy efficiency. The table also shows that THEX3 hybrid had the lowest energy requirement while THEX1 recorded the least lost work and highest exergetic efficiency of 92.7 % followed by THEX3.

Table 3 shows the result of exergetic efficiency and lost work for the hybrid configurations. Meanwhile, Tables 4 and 5 shows the exergy efficiency across the constituent equipment/devices of the hybrids. As it is expected, the results show that the higher the aggregate exergetic efficiency across the units the smaller the lost work (irreversibility). These values indicate the work potentials of the material or energy stream. Considering THEX1 which is the most energy efficient hybrid, the exergetic efficiency of most of the equipment in the flow sheet were around 99 % with the exception of valve, stripper (C1) and rectifier (C2). This trend was observed with the other five hybrids. This indicates a greater need for improved equipment design in the THIDC section.

The least energy efficient hybrid, THAZ2, was observed to have the highest exergy losses at the extractive column, recovery column (C3), heat exchanger E2 and valve as shown in Table 4. Selection of heat transfer media possessing excellent thermal and transport properties should improve efficiency of heat exchanger.

The loss of exergy in mixers and columns, particularly in the dehydration unit of most of the hybrids is mainly due to the change in composition from inlet to the outlet stream.

This loss resulting from exergy change of mixing can be addressed by blending as much as possible streams of nearly similar compositions. The low exergetic efficiency of some columns is due to inevitable change in composition across these units.

Similar values were across the hybrids investigated with most of the equipment of THIDC unit. This was more prominent in the hybrids containing extractive sequences. The reason for this has to do with similar operating conditions used with the THIDC equipment and auxiliaries. Meanwhile, it is different for hybrid containing azeotropic distillation units because of the slight interactions between THIDC and the dehydration unit.

A look at the two most energy efficient hybrids (THEX1 and THEX3), brought to the fore the possibility that distillation column may have high exergy loss but very high or low exergetic efficiency as pointed out earlier [25]. THEX3 has higher exergy loss and lower energy requirement than THEX1 yet the latter is exergetically more efficient. The results also show that a process requiring less energy is not necessarily more efficient than the one with greater energy input if the former process is more efficient in its energy utilization.

Table 4. Equipment Exergy Efficiency for Azeotropic Hybrids.

Equipment	% Equipment Exergy Efficiency		
	THAZ1	THAZ2	THAZ3
Stripper (C1)	97.64	97.64	97.33
E1	99.99	99.99	99.99
Compressor	99.50	99.50	99.48
Rectifier (C2)	93.54	93.43	93.74
Valve	93.24	93.24	93.35
E2	91.64	91.64	91.64
Mixer	99.53	99.96	99.87
Column C3	88.50	47.74	81.50
Column C4	54.00	39.21	96.91
Splitter/Mixer	-	99.98	99.96
E3-condenser	-	-	98.80
Decanter	-	-	99.96

Table 5. Equipment Exergy Efficiency for Extractive Hybrids.

Equipment	% Equipment Exergy Efficiency		
	THEX1	THEX2	THEX3
Stripper (C1)	97.65	97.65	97.65
E1	99.99	99.99	99.99
Compressor	99.51	99.51	99.51
Rectifier (C2)	93.65	93.64	93.64
Valve	95.81	95.81	95.81
E2	99.98	99.98	99.98
E3	99.99	99.98	99.98
Column C3	99.49	99.55	99.27
Column C4	99.81	96.57	95.15

THIDC-extractive distillation hybrids recorded the least energy consumption and lost work among all the hybrids. The THEX1 hybrid is the most energy efficient followed by comparison with THEX2 and THEX3 hybrid in that order. Though, the energy consumption and lost work did not differ much among the three, but THEX3 is the least energy requiring. Previous works that used conventional column with the same extractive sequences but identified hybrid with Petlyuk column as the most energy efficient and least energy consuming [9]. Comparing the results in this work and other work [9] that used conventional distillation column (as part of the hybrid) showed energy cost savings of 70, 71 and 58.3 % in processing a kilogram of the feed using THEX1, THEX2 and THEX3 respectively (see Table 6). The improvement recorded in this work can also be linked to the use of THIDC.

The exergy analysis has placed THIDC-Conventional extractive distillation hybrid as the best hybrid. It has the least lost work (irreversibility rate) and highest exergy efficiency followed by THEX3 and THEX2 in that order. The least in term of thermodynamic potential are the hybrids with azeotropic dehydration units.

Comparative analysis based on energy consumption from input utility streams show that a reasonable improvement was achieved in this work. In comparison to a previous

Table 6. Comparative Energy Consumption (Present and Previous Studies)

Current Studies		Previous Studies			
Hybrid Configuration	Energy Consumed (MJ/kg)	Hybrid Configuration	Energy Consumed (MJ/kg)	% Energy cost savings	Basis
THEX 1	0.56	Conventional distillation – EX1 [9]	1.86	70	Feed
THEX 2	0.56	Conventional distillation – EX2 [9]	1.89	71	Feed
THEX 3	0.57	Conventional distillation – EX3 [9]	1.38	58.3	Feed
THEX 1	5.44	Conventional distillation – EX1 [8]	10.69	49.1	Product
THAZ 1	9.14	Conventional distillation – AZ1 [8]	12.38	32.4	Product

[8], Table 6 shows that there were 49.1 and 32.4 % cost savings respectively in using THEX1 and THAZ 1 hybrid to produce a kilogram of 99.5 wt% anhydrous fuel ethanol. The reduction in energy consumption achieved in all cases could be attributed to the use of THIDC which has been found to be relatively efficient distillation column type [26, 19]. This column used higher grade energy (work) to achieve desired separation enrichment of bioethanol mash to azeotropic mixture. However, the study of the feed thermal condition allowed the choice of optimal THIDC feed condition [19]. This might be another reason for the observed improvement.

### 3.2 Economic Analysis

Figure 7 shows the various cost elements of each of the hybrids. Though, the feed and desired product specifications were the same but side products, recycle streams and entrainer specifications were not. This disparity could explain, in part, the variation in total annual products sales (TAPS) among the hybrids.

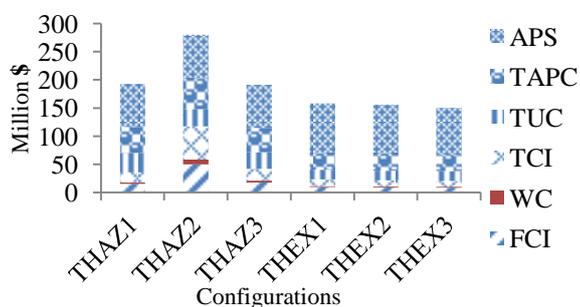


Figure 7. Product Sales and Cost Analysis Result of Hybrid Configurations

The THIDC-azeotropic sequences recorded the least total annual product sales (TAPS) of only \$ 76.91 million because the by-product value was less. The hybrid with the highest TAPS was THEX1 with \$ 87 million. THAZ3 had the least annual product cost of \$ 29.4 million followed by THAZ1 and THAZ2 in that order. THAZ2 had the highest annual product cost.

Profitability analysis shows that THIDC-extractive sequences hybrids are the most attractive configurations for the refining of bioethanol to fuel grade (see Figures 8 to 11). The profitability analysis showed THEX 3 as the most favorable. It had the highest return on investment (ROI), maximum annual and continuous discounted cash flow rate (ADCR and CDCR) of 114 %/year and 76.1 %/year respectively (Figure 8). It also has the least payback period (see Figure 10). The second and third best were THEX1 and THEX2 respectively. THAZ2 is the least economic hybrid configuration of the six. The net present worth was in favor of THEX1 but THEX3 was the second best.

The ultimate economic requirement of this kind of process is the maximization of profit and minimization of product cost. It is therefore, imperative to use return on investment as an economic criteria for comparison among all hybrid configurations.

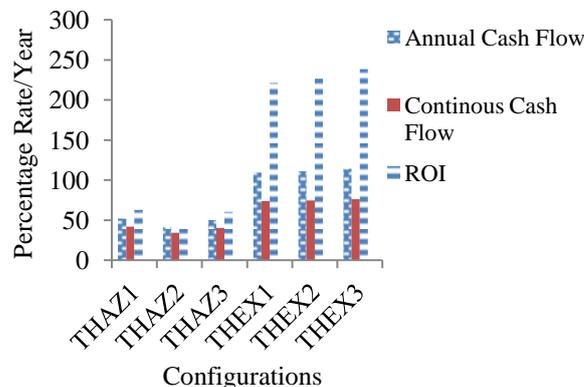


Figure 8. Return on Investment and Cash flow Rate for Hybrid Configurations

It is necessary to consider a trade-off between exergy and economic efficiency to draw conclusion as to which hybrid configuration is the most efficient. This is because the most exegergetically efficient hybrid may be the most economically efficient. Looking at both sides, economic consideration takes precedence because of smaller difference of 0.9 % in the exergetic efficiency between highest exergetic and most economic hybrid.

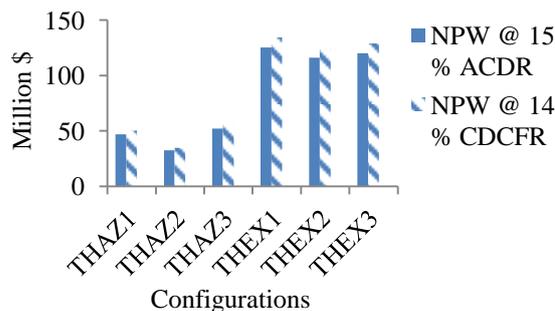


Figure 9. Hybrid Configurations Net Present worth

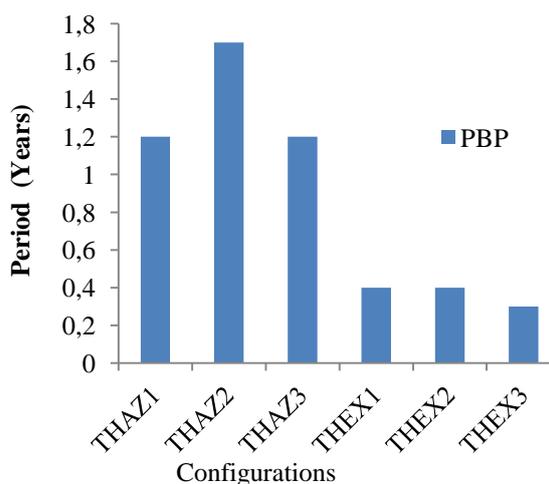


Figure 10. Hybrid Configurations Pay Back Period

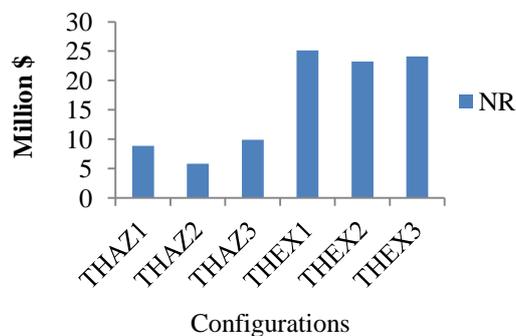


Figure 11. Hybrid Configurations Net Return

Looking at energy, exergy and economic analysis together, THIDC-Thermally coupled extractive distillation hybrid is the best hybrid configuration of the six distillation hybrid configurations studied for bioethanol refining. This is because the hybrid was favoured from economic analysis and its energy consumption and irreversibility rate were very much close to the minimum obtainable among all the hybrid configurations.

#### 4. Conclusions

The THIDC-extraction sequences were found to be better than the azeotropic distillation derived hybrids thermodynamically and economically. THEX1 recorded the highest exergy efficiency while THAZ2 was identified as the least efficient configuration.

It was also observed that the less energy consuming process might not necessarily be the most efficient configuration. However, economic analysis suggested the less energy consuming process THEX3 as the most attractive configuration that guarantee the production of affordable and sustainable fuel grade bioethanol.

Thermally coupled extractive distillation and THIDC hybrid is the preferred choice for commercial refining of bioethanol mash to fuel grade among all hybrid configurations considered in this work. Savings in energy cost of as much as 71 % was achieved for THEX hybrids over similar configurations used in an earlier work. Utility energy input cost improvement was as high as 49 and 32 % for THEX and THAZ hybrids respectively, when compared to reported works.

#### Nomenclature

APS	Annual product sales
AZ1	Azeotropic distillation with recycle stream mixing and entering with organic phase
AZ2	Azeotropic distillation with recycle stream mixing and entering with feed
AZ3	THIDC-Azeotropic distillation with recycle
CDCFR	Continuous discounted cash flow rate
ADCFR	Annual discounted cash flow rate
CEPCI	Chemical engineering cost index
Conv.	Conventional
EX1	Conventional Extractive Distillation
EX2	Petlyuk Extractive Distillation
EX3	Thermally Coupled Extractive Distillation
FCI	Fixed Capital Investment
NR	Net Return
NPW	Net Present Worth
PBP	Payback Period
ROI	Return on Investment

TAPC	Total annual product cost
TCI	Total capital investment
THAZ1	THIDC-Azeotropic distillation with recycle stream
THAZ2	THIDC-Azeotropic distillation with recycle stream mixing and entering with feed
THAZ3	THIDC-Azeotropic distillation with recycle stream entering into decanter
THEX1	THIDC-Conventional extractive distillation
THEX2	THIDC-Petlyuk extractive distillation
THEX3	THIDC Thermally Extractive Distillation
THIDC	Totally heat integrated distillation column
TUC	Total annual utility cost
WC	Working capital

#### References

- [1] Bastidas, P.A., Gil, I.D. and Rodriguez, G. (2010). Comparison of the Main Ethanol Dehydration Technologies through Process Simulations. In *20th European Symposium on Computer Aided Process Engineering-ESCAPE-20*, Ischia, Italy.
- [2] Vascncelos, C.J.G., Wolf-Maciel, M.R. (2002). Optimization, Dynamics and Control of a Complete Azeotropic Distillation: New Strategies and stability Consideration. *Distillation and Absorption Conference, Baden-Baden* 6-28, 2002
- [3] Watanabe, D., Wu, H., Noguchi, C., Zhou, Y., Akao, T., and Shionoi, H. (2010). Enhancement of the initial rate of ethanol fermentation due to dysfunction of yeast stress response components Msn2p and/or Msn4p. *J. Applied and Environmental microbiology*, 77, 934-941.
- [4] Jeong, J., Jeon, H., Ko, K., Chung, B., and Choi, G. (2012). Production of Anhydrous Ethanol Using Various Pressure Swing Adsorption Processes in Pilot Plant. *Renewable Energy J.*, 42, 41-45.
- [5] Demirel, Y. (2004). Thermodynamics Analysis of Separation Systems. *J. Separation Science and Technology*, 39, 3897-3942.
- [6] Bremers, G., Birzietis, G., Blija, A., Skele, A., Rucins, A. and Danilevics, A. (2010). Evaluating Usability of Water Adsorption and Rectification in Dehydration of Bioethanol. *Jelgava 2010: Proceedings of 9th International Scientific Conference of Engineering for Rural development*, 154-157.
- [7] Gil, I.D., Uyazan, A.M., Aguilar, J.L., Rodriguez, G., and Caicedo, L.A. Simulation of Ethanol Extractive Distillation with a Glycols Mixture as Entrainer. In *2nd Mercosur Congress on Chemical Engineering & 4th Mercosur Congress on Process Systems Engineering*, Braga, Portugal, September 4-6, 2008.
- [8] Sanchez, O.J., Moncada, J.A., and Cardona, C.A. (2006). Modeling and Simulation of Ethanol Dehydration by Pervaporation and Energy Analysis of Separation Schemas for Fuel Ethanol Production. In *Int. Congress of Chemical and Process Engineering*
- [9] Torres-Ortega, C.E., Segovia-Hernandez, J.G., Harnandez, S., Hernandez, H., Bonilla-Petriciolet A.B. and Maya-Yescas, R. Design and Optimization of Thermally Coupled Distillation Sequences for Purification of Bio-ethanol. *PSE2009: In 10th*

*International Symposium on Process Systems Engineering*, 957-962. Salvador-Bahia, Brazil August 16-20, 2009.

- [10] Bremers, G., Birzietis, G., Blija, A., and Danilevics, A., and Skele, A. Scheme of Technology for Congruent Dehydration of Bioethanol in Semi-Dry Way. *2011: Proceedings of 10<sup>th</sup> International Scientific Conference of Engineering for Rural development*, 244-247. Jelgava 2011
- [11] Carmo, M.J. and Gubulin, J.C. (1997). Ethanol Water Adsorption on Commercial 3A zeolites: Kinetic and Thermodynamic data. *Brazilian Journal of Chemical Engineering*, 14, 1-17.
- [12] Pruksathorn, P. and Vitidsant, T. (2009). Production of Pure Ethanol from Azeotropic Solution by Pressure Swing Adsorption. *American J. Engineering and Applied Sciences*, 2, 1-7.
- [13] Sohel, M.I. and Jack, M.W. Thermodynamic Analysis and Potential Improvements of a Biochemical Process for Lignocellulosic Biofuel Production. In *World Renewable Energy Congress*, Linköping, Sweden, 8-13 May 2011.
- [14] Naidu Y. and Malik R.K. (2011). Generalized Methodology for Optimal Configurations of Hybrid Distillation-Pervaporation Processes. *J. Chemical Engineering Research and Design*; 89, 1348-1361.
- [15] Verhoef A., Degreve J., Huybrechs B., Veen H., Pex P and Bruggn B.V. (2008). Simulation of a Hybrid Pervaporation-Distillation Process. *J. Computers and Chemical Engineering*, 32, 1135-1146.
- [16] Kunnakorn D., Rirksomboon T., Siemanond K., Aungkavattana P., Kuanchertchoo N., Chuntanalerg P., Hemra K., Kulprathipanja S., James R.B, Wongkasemjit S. (2013). Techno-economic comparison of energy usage between azeotropic distillation and hybrid system for water-ethanol separation. *Renewable Energy*, 51, 310-316.
- [17] Salvador H. (2008). Analysis of Energy-Efficient Complex Distillation Options to Purify Bioethanol, *Chem. Eng. Technology*, 31(4), 597-603.
- [18] Errico M., Giuseppe T., Ben-Guang R., Daniele D., Ilkka T. (2009). Energy Saving and Capital Cost Evaluation in Distillation Column Sequences with a Divided Wall Column. *Chemical Engineering Research and Design*, 87, 1649-1657.
- [19] Suleiman, B. Exergetic-economic assessment of Bioethanol Refining Configurations. MSc. Thesis Unpublished, Department of Chemical Engineering, Ahmadu Bello University, Zaria, Nigeria, 2014.
- [20] Neagu M., Cursaru D. (2013). Bioethanol dehydration by Extractive Distillation with Propylene Glycol Entrainer (A preliminary case study). *REV. CHIM (Bucharest)*, 64, 92-94.
- [21] Querel, E., Gonzalaz-Reguerel, B., and Perez-Benedito, J.L. (2013). *Practical Approach to Energy and Thermo-economic Analysis of Industrial Processes*, (1<sup>st</sup> ed.). London Heidelberg: Springer London.
- [22] Seader J.D., Henley E.J. and Roper D.K. *Separation Process Principles*, (3<sup>rd</sup> Ed.). Hoboken United State of America: John Wiley and Sons. Inc., 2011
- [23] Dincer, I. and Naterer G.F. (2010). Assessment of Exergy Efficiency and Sustainability Index of an Air-Water Heat Pump. *International J. Exergy*, 7, 37-50.
- [24] Peters M.S., Timmerhaus K.D. *Plant Design and Economics for Chemical Engineers* (4<sup>th</sup> ed.). New York: McGraw-Hill Inc., 1991.
- [25] Bandyopadhyay, S. (2002). Effect of Feed on Optimal Thermodynamic Performance of a Distillation Column. *J Chem. Eng.*, 88, 175-186.
- [26] Wolf Maciel, M.R., (2009), *Advances in ethanol purification: alternatives and perspectives*. Retrieved from [http://www.fapesp.br/eventos/2009/09/10\\_bioen\\_Regina.pdf](http://www.fapesp.br/eventos/2009/09/10_bioen_Regina.pdf)
- [27] Online process equipment calculator. Peters and Timmerhaus (1991). Retrieved June 12, 2012 from [http://www.higheredmcgrawhill.com/sites/0072392665/student\\_view0/cost\\_estimator.html](http://www.higheredmcgrawhill.com/sites/0072392665/student_view0/cost_estimator.html)