### **Research article**

Aliyu F., et al., Isolation and Characterisation of the Degradation Potential of 2,2-Dichloropropionate (2,2-Dcp) By Bacillus amyloliquefaciens From Gebeng. International Journal of Life Sciences and Biotechnology, 2022. 5(2). p. 200-212. DOI: 10.38001/ijlsb.1064391

# Isolation and Characterisation of the Degradation Potential of 2,2-Dichloropropionate (2,2-Dcp) By *Bacillus amyloliquefaciens* From Gebeng

Firdausi Aliyu<sup>1,2</sup>, Aliyu Adamu<sup>1,3</sup>, Roswanira Abdul Wahab<sup>4</sup>, Fahrul Huyop<sup>1</sup>

#### ABSTRACT

Most halogenated compounds such as  $\alpha$ -halocarboxylic acid ( $\alpha$ HCA) are potential carcinogens that are toxic and widely liberated in both aquatic and terrestrial ecosystem. 2,2-dichloropropionate (2,2-DCP) is a synthetic halogenated compound commonly used in agricultural activities as an herbicide. This study is focused on the isolation and characterization of dehalogenase producing bacteria capable of utilizing 2,2-dichloropropionate as the sole carbon and energy source from soil sample obtained from the industrial hub of Gebeng (Pahang), Malaysia. An enrichment culture technique supplemented with 2,2-DCP was used to isolate the bacteria. A pure culture strain designated as strain FDS grew well in media containing 20 mM 2,2-DCP at a temperature of 30°C exhibiting a cell doubling time of 22.21hours and a maximum rate of chloride ion release of 0.345mmol/L suggesting this is the optimum concentration for growth. Based on microscopic observations, biochemical characteristics and phylogenetic analysis of the 16S rRNA gene sequence, FDS was identified as *Bacillus amyloliquefaciens* with 98% homology. A literature survey revealed no report regarding the role of *B. amyloliquefaciens* on pesticide biodegradation as well as on 2,2-DCP, therefore the current isolate was designated *Bacillus amyloliquefaciens* FDS.

#### ARTICLE HISTORY Received

28 January 2022 Accepted 29 April 2022

#### **KEYWORDS**

Dehalogenase, α-halocarboxylic acid, 16S rRNA, Biodegradation, 2,2-DCP

# Introduction

The advancement in technology has resulted in the formation of various man-made organic compounds. These compounds (xenobiotics) are produced in enormous quantities which also benefit our life. Among these xenobiotics are halogenated compounds. These

<sup>&</sup>lt;sup>1</sup> Department of Biotechnology, Faculty of Biosciences and Bioengineering, Universiti Teknologi Malaysia, 81310 Johor Bahru, Malaysia

<sup>&</sup>lt;sup>2</sup> Department of Biotechnology, Faculty of Science, Nigerian Defence Academy (NDA),Kaduna PMB2109,Nigeria

<sup>&</sup>lt;sup>3</sup> Department of Microbiology, Faculty of Science, Kaduna State University, Tafawa Balewa way, Kaduna PMB 2339, Nigeria.

<sup>&</sup>lt;sup>4</sup> Department of Chemistry, Faculty of Science, Universiti Teknologi Malaysia, 81310 UTM Johor Bahru, Malaysia.

<sup>\*</sup>Corresponding author: Firdausi Aliyu, afirdausi@nda.edu.ng

compounds are extensively used as antibiotics (chloramphenicol, vancomycin), anticancer (iododoxorubicin, salinosporamide) and as agrochemicals (herbicides, pesticides, soil fumigants) and solvents (lubricants, silicones) [1]. Halogenated compounds have been reported to cause serious health issues such as liver necrosis, dissociation of hepatocytes, fat disintegration [2], hormonal imbalance, lower respiratory problems, skin and lung cancer etc. [3], in addition to causing serious environmental pollution owing to their toxicity, their toxic breakdown products, persistence and bioaccumulation in the food chain and water bodies [4]. These compounds are poisonous and very difficult to degrade [5].

2,2 -Dichloropropionate (2,2-DCP) is a chlorinated aliphatic acid also known as Dalapon. It is used in agriculture as an herbicide and a plant growth regulator used on certain annual and perennial grasses. It kills only certain plants while sparring non-target plants thereby preventing lipid synthesis, stops cell division, reduces wax production by leaves, affects carbohydrate, lipid and nitrogen metabolism [6].

Microorganisms play an important role in the degradation of xenobiotics, they help transform toxic compounds into non-hazardous substances. Some groups of bacteria use halogenated organic compounds as their main carbon and energy source, these microorganisms detoxify or remove pollutants owing to their diverse metabolic capabilities thereby helping reverse the effect of environmental halogen- associated pollution [7].

These bacteria produce enzymes known as dehalogenase which catalyze the cleavage of carbon-halogen bonds in halogenated organic compounds thereby producing some environmental friendly products by a process known as dehalogenation [8], [9]. Some examples of aerobic degradative bacteria include; *Pseudomonas, Bacillus, Micrococcus* and *Rhodococcus* while anaerobic degradative bacteria are *Syntrophus, Methanospirillum, Syntrophobacter*. Among them *Pseudomonas* and *Bacillus* species are the most studied [10], [11]. The present study is focused on the isolation and characterization of 2,2-DCP degrading bacteria from Gebeng industrial area using enrichment techniques.

201

# **Materials and Methods**

## Isolation and identification of 2,2-DCP degrading bacteria

The soil sample was collected from the industrial area of Gebeng (Pahang), Malaysia. The industries around the are generate a lot of chemical waste containing halogen, which contaminate soil in the environment.

## **Preparation of medium for isolation**

Varying concentrations (10 mM, 20 mM, 30 mM and 40 mM) of 2,2-DCP liquid minimal medium were prepared according to Table 1. tenfold of trace metal salts which comprised of Nitriloacetic acid (1.0g/L), MgSO<sub>4</sub>.7H<sub>2</sub>O (2.0g/L), FeSO<sub>4</sub>.7H<sub>2</sub>O (0.12g/L), MnSO<sub>4</sub>.4H<sub>2</sub>O (0.03g/L), CoCl<sub>2</sub>.6H<sub>2</sub>O (0.01g/L), ZnSO<sub>4</sub>.H<sub>2</sub>O (0.03g/L) were dissolved in distilled water. The basal salts were prepared as a tenfold concentration of K<sub>2</sub>HPO<sub>4</sub>.3H<sub>2</sub>O (42.5g/L), NaH<sub>2</sub>PO4.2H<sub>2</sub>O (10.0g/L), (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> (20.0g/L) in distilled water. The medium for isolation contained 10ml of 10x trace metal salts and 10ml of 10x basal salts per 100ml distilled water were autoclaved (121°C, for 15min at 15psi). Then 2ml of 2,2-DCP (1M) was filter sterilized using a nylon filter (0.2µm pore size) and added aseptically into the media and made up to the desired final concentration the plates were prepared by adding bacteriological agar (1.5% w/v) prior to sterilization[12].

2,2-DCP concentration	10mM	20mM	30mM	40mM
Distilled water	79	78	77	76
Trace metal	10	10	10	10
Basal Salt	10	10	10	10
V1, 1M	1	2	3	4
Total (ml)	100	100	100	100

Table1 Composition of minimal media

### **Isolation procedure**

5g of sample was added into a 250ml shaker flask containing 100ml minimal salts and supplemented with 20 Mm 2,2-DCP as the sole carbon and energy source. Following 10

days incubation with shaking at 200rpm at 37°C, an aliquot (0.1ml) was pipette out and spread on plates supplemented with 20 Mm 2,2-DCP and incubated at 37°C. The resulting colonies were selected and screened. Pure cultures, the resulting colonies were repeatedly sub cultured in the same medium to confirm carbon source utilizing ability.

#### **Identification of isolate**

To ascertain the identity of the isolate, Gram staining, motility, spore staining and biochemical test were carried out: Oxidase, catalase, gelatin liquefaction, lactose fermentation, citrate test, Nitrate reduction, indole, urease and starch hydrolysis test [12].

## **Growth profile**

The growth profile of the bacterial culture in different concentrations of 2,2-DCP (10mM-40mM) were monitored by taking 1ml of the sample from the growth medium at 2 hours interval over 20 hours growth period and measured at A<sub>680nm</sub> using Jenway 6305 UV Spectrophotometer (Staffordshire, UK).

#### Halide ion assay

The utilization of 2,2-DCP by the isolates results in the release of halide ion during dehalogenation reaction as carried out from adaptation of [13]. A sample (1ml) from the liquid growth medium, 100 $\mu$ l of Reagent I (0.25M ferric ammonium sulfate in 9M nitric acid) was added and thoroughly mixed. To this, Reagent II (100 $\mu$ l of mercuric thiocyanate dissolved in saturated ethanol) was also added and the solution was mixed by shaking. Color could develop for 10 minutes and measured at A<sub>460nm</sub>.

#### **16S rRNA genomic analysis**

The genomic DNA was extracted from an overnight bacterial culture using Promega® Wizard® Genomic DNA Purification Kit (Promega,Madison, WI, USA). The purification and merging the DNA concentration was measured by Nanodrop machine.The Polymerase chain reaction (PCR) was carried out using universal primers, Fd1 (5'-AGA GTT TGA TCC TGG CTC AG-3') and rP1 (5'-ACG GTC ATA CCT TGT TAC GAC TT-3') to amplify the DNA fragment as suggested by [14], [15]. The PCR cycle for DNA amplification was performed for 30 cycles and set as initial denaturation at 95°C for 5 minutes; cooling, denaturation at 94°C for 1 minute; annealing at 55°C for 1 minute; final

elongation at 72°C for 10 minutes. The PCR products were purified using QIA quick PCR purification kit and sequenced by 1<sup>ST</sup> Base Laboratory, Malaysia.

## Phylogenetic study

The bacterial 16S rRNA gene sequences were compared with other sequences of dehalogenase producing bacteria in GenBank database using BLASTn- The FASTA format of all 16S rRNA sequences obtained were aligned together by MUSCLE using MEGA7 [16]. The topology of the tree was constructed using neighbor-joining method with bootstrap test values above 50% based on 500 resampling.

# Results

## Isolation of 2,2-DCP degrading bacteria

Following 5 days incubation of the soil sample obtained from Gebeng industrial area lead to the isolation of a bacterium designated strain FDS. The Size, morphology (Table 2) while Gram staining showed the microorganism to be Gram positive rod shape (figure 1). However, to fully ascertain the identity of strain FDS, biochemical tests (Table 3) were performed according to Bergy's Manual of Systematic Bacteriology [17].

Parameters	Results
Pigmentation (pure colony)	Creamy
Elevation	Flat
Colony	Rough circular
Margin of the colony	Undulate
Microscopic shape	Rod
Size	3.40µm
Spore formation	Negative
Oxygen requirement	Aerobic

Table 2 Morphological characteristics of Strain FDS on solid media



Fig 1 Microscopic observation showing the Gram-stain of strain FDS (1000 X magnification

Test	Results
Oxidase	Positive
Catalase	Positive
Indole	Positive
Gelatin liquefaction	Positive
Motility	Negative
Citrate	Negative
Urease	Positive
Nitrate reduction	Positive
Starch hydrolysis	Positive
Lactose utilization	Negative

Table 3 Results from the Partial biochemical tests

## Growth profile and Halide ion assay

The growth profile of FDS in different concentrations (10mM-40mM) of 2,2-DCP are shown (Figure2). 20mM 2,2-DCP exhibited good bacterial growth. 10mM and 30mM 2,2-DCP showed low growth, while 40mM 2,2-DCP exhibited no growth which suggests that the concentration was toxic. Following this observation, the bacterium FDS was grown in

liquid minimal medium supplemented with 20mM 2,2-DCP at 37°C with agitation at 200 rpm with growth measured at 2 hours interval over 20 hours period. The doubling time of the bacterium was 22.21 hours with a maximum chloride ion of 0.345mmol/L.



Fig 2 Growth rate of strain FDS in different concentrations of 2, 2-DCP

### Bacterial growth on other halogenated compounds

The bacterium FDS was grow in 10mM trichloroacetic acid (TCA), 10mM 2chloropropionate (2-CP) and 10mM 2-bromopropionate (2-BP) liquid minimal medium. The results (Table 4) showed growth in 10mM TCA while no growth was observed in 2-CP and 2-BP which suggests the substrate is not suitable for growth. The cell doubling time of FDS grown in 10mM TCA was 14.09 hours and a maximum chloride ion of 0.223mmol/L (Figure 3).

<b>Fable 4</b> Growth properties	of strain FDS	S in different	substrate
----------------------------------	---------------	----------------	-----------

Growth Substrate (10mM)	Maximum Medium Turbidity (A680 <sub>nm</sub> )	Doubling time(hours)
Trichloroacetic acid	0.184	14.09
2-bromopropionate	NG*	NG*
2-chloropropionate	NG*	NG*

<sup>\*</sup>NG: No Growth



Fig 3 Growth rate of strain FDS in 10mM concentration of TCA

## Identification of 2,2-DCP degrading bacteria using 16S rRNA analysis

The genomic DNA of FDS was successfully extracted and amplified using universal primers Fd1 and rP1. Gel electrophoresis of the PCR product revealed a single fragment of approximately 1500bp (Figure 4).



Fig 4 PCR product for 16S rRNA under UV light

Lane 1: 1kb DNA ladder, Lane 2: PCR product of 16S rRNA, Lane 3: Control ( $dH_2O + Fd1 + PCR mix$ ), Lane 4: Control ( $dH_2O + rP1 + PCR mix$ ).

## Sequencing and analysis of 16S rRNA gene

The outcome of the PCR product sequenced by 1<sup>st</sup> Base Company was received in ". abl" format. The files were viewed using Sequence scanner software. The BLASTn results showed that bacterium FDS shared 98% identity to *Bacillus amyloliquefaciens* (Table 5).

Accession number	Description	Maximum score	Query coverage (%)	Identity (%)
NC 014551.1	Bacillus amyloliqueficiens DSM7	1995	97	98
NC 000964.3	<i>Bacillus subtilis</i> strain 168	1973	97	98
NZ CP007640.1	<i>Bacillus atrophaeus</i> subsp. globiqii strain BSS	1962	97	97

 Table 5 The top ten BLAST search result of strain FDS

# Phylogenetic study

The top dehalogenase producing bacteria obtained from BLASTn analysis as obtained from NCBI database were selected for the construction of phylogenetic tree using MEGA 7 by neighbour-joining with bootstrap value (Figure 5). The results inferred that the bacterium FDS may be designated as *Bacillus amyloliquefaciens* FDS.



Fig 5 Phylogenetic relationships between *Bacillus amyloliquefaciens* FDS and other dehalogenase producing bacteria16S rRNA sequences

## Discussion

In this study, Soil sample obtained from Gebeng industrial area which is synonymous with the manufacturing of various chemical and petrochemical products was used as the source for the isolation of bacteria capable of degrading 2,2-DCP and TCA as the sole carbon and energy. The isolation procedure employed in this study entailed the enrichment of minimal salts with the carbon sources (2,2-DCP and TCA), followed by plating on a similar agar medium. The two procedures help ensures that the isolation is that of bacteria only capable of utilizing the carbon sources. The successful isolation of a strain designated FDS was observed to have various morphological (Table 4.1) and biochemical characteristics (Table 4.2). However, to properly characterize and identify the strain a method of bacterial taxonomy known as 16S rRNA was employed this routine procedure in prokaryotic taxonomy results in large and growing databases which improve identification results, phylogeny reconstructions and primer specificity evaluations[18]. The 16S rRNA gene sequence analysis suggests that strain FDS gene sequence shared at least 98% identity to the sequence of Bacillus amyloliquefaciens which suggests strain FDS belongs to the Bacillus sp. (Figure 4.9). Thus the 2, 2-dichloropropionate degrading bacteria isolated belong to Bacillus sp. This was further supported by staining properties and biochemical analysis and in accordance with Bergy's Manual of Systematic Bacteriology [17]. Bacillus *amyloliquefaciens* is a gram positive, rod-shape, catalase positive and aerobic soil bacteria. As with other members of the *Bacillaceae* they have the ability to form endospore allowing them to survive for a prolonged period time, thereby increasing their soil shelf life [19], [20]. Previous studies have shown that several strains of the bacteria belonging to the genus Bacillus particularly B.subtilis and B.amyloliquefaciens species were known to be effective for the biocontrol of many plant-diseases caused by soil borne pathogens [21], [22] or postharvest pathogens [23]–[25]. Their ability to colonize plants rhizosphere thereby suppressing competing phytopathogenic bacteria and fungi these classifies them as microbial biopesticides. They represent about half of the commercially available bacterial biocontrol agents [26]. However, there has not been any report of Bacillus amyloliquefaciens that is able to degrade 2,2-dichloropropionate and trichloroacetate reported so far. Therefore, this strain of *Bacillus amyloliquefaciens* that can degrade 2,2dichloropropionate deserves more research.

They are four basic criteria for a given halogenated compound to be utilized by an organism as a sole carbon and energy source as reported by [27]. Firstly, the organism should be able to synthesize enzyme (dehalogenase) which can remove the substituent halogen(s) from the compound. Secondly, the product of dehalogenation should be nontoxic and easily converted to an intermediate which can readily be absorbed in the organism's central metabolic pathway. Thirdly, entry of halogenated compound into the organism should be either through passive or active transport in order for the compound to be synthesized by the enzyme, and finally, the halogenated compound should not be toxic to the organism at intracellular concentrations [27]. These criteria were satisfied when Bacillus amyloliquefaciens FDS grew on 2,2-dichloropropionate and trichloroacetic acid as the sole carbon and energy source. The isolate grew more rapidly on 2,2-dichloropropionate compared to trichloroacetic acid, possibly because the organism was originally isolated using 2.2-dichloropropionate. 2.2-dichloropropionate failed to support growth at concentration exceeding 40mM, while trichloroacetic acid failed to support growth exceeding 10Mm; presumably the intracellular concentration of halo-aliphatic acid had reached a toxic level.

## Conclusion

The successful isolation identification and characterization of *Bacillus amyloliquefaciens* FDS capable of degrading 2,2-DCP and TCA from Gebeng would suggest that these study areas possess enormous reservoir of unexplored/unexploited microbes. In conclusion, this study provides the identity of a bacterium that can utilize halogenated compound and used in bioremediation strategies. In future, the study can help describe the diversity of microbial dehalogenases and the dehalogenation enzyme regulation.

#### Abbreviations

<sup>2,2-</sup>DCP: 2,2-dichloropropionic acid, BLAST: Basic Local Alignment Search Tool, Br: Bromine, Cl: Chloride, DNA: Deoxyribonucleic acid, EDTA: Ethylenediaminetetraacetic acid, EtBr: Ethidium Bromide, F: Flourine, I: Iodine, LB: Luria Bertani, MEGA: Molecular Evolutionary Genetics Analysis, NaCl: Sodium chloride, NADH: Nicotinamide adenine dinucleotide, NCBI: National Centre for Biotechnology Information, PCR: Polymerase chain reaction, RNA: Ribonucleic acid, sp.: Species, TAE: Tris acetate EDTA, TCA: Trichloroacetic acid, UV: Ultraviolet

#### **Conflict of Interest**

The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

### References

- 1 A. Adamu, R. A. Wahab, and F. Huyop, "l-2-Haloacid dehalogenase (DehL) from Rhizobium sp. RC1.," Springerplus, 2016. 5(1): p. 695.
- 2 M. Sathishkumar, A. R. Binupriya, S. H. Baik, and S. E. Yun, "Biodegradation of crude oil by individual bacterial strains and a mixed bacterial consortium isolated from hydrocarbon contaminated areas," CLEAN–Soil, Air, Water, 2008. 36(1): p. 92–96.
- F. Aguilera, J. Méndez, E. Pásaro, and B. Laffon, "Review on the effects of exposure to spilled oils on human health," Journal of Applied Toxicology, 2010. 30 (4): p. 291–301.
- 4 F. Huyop and M. Nemati, "Properties of dehalogenase from Rhizobium sp," RC1. African Journal of Microbiology Research, 2010. 4 (25) : p. 2836–2847.
- 5 Y. Wang, Y. Xin, X. Cao, and S. Xue, "Enhancement of L-2-haloacid dehalogenase expression in Pseudomonas stutzeri DEH138 based on the different substrate specificity between dehalogenaseproducing bacteria and their dehalogenases," World Journal of Microbiology and Biotechnology, 2015. 31(4): p. 669–673.
- 6 F. M. Ashton and A. S. Crafts. 1973, Mode of action of herbicides. Mode of action of herbicides.
- J. I. Medina-Bellver et al., "Evidence for in situ crude oil biodegradation after the Prestige oil spill," Environmental Microbiology, 2005. 7(6): p. 773–779.
- D. Alomar et al., "Molecular characterization of monochloroacetate-degrading Arthrobacter sp. Strain d2 isolated from Universiti Teknologi Malaysia agricultural area," Bioremediation Journal, 2014.18 (1): p. 12–19.
- 9 M. F. Edbeib, R. A. Wahab, F. Z. Huyop, H. M. Aksoy, and Y. Kaya, "Further analysis of Burkholderia pseudomallei MF2 and Identification of putative dehalogenase gene by PCR," Indonesian Journal of Chemistry, 2020. 20 (2): p. 386–394.
- 10 S. Chowdhury, M. Mishra, V. Adarsh, A. Mukherjee, A. R. Thakur, and S. R. Chaudhuri, "Novel metal accumulator and protease secretor microbes from East Calcutta Wetland," American Journal of Biochemistry and Biotechnology, 2008. 3: p. 255–264.
- 11 Y. Varsha, C. Naga Deepthi, and S. Chenna, "An emphasis on xenobiotic degradation in environmental clean up," Journal of Bioremediation and Biodegradation, 2011. 11: p. 1–10.
- 12 W. A. Hareland, R. L. Crawford, P. J. Chapman, and S. Dagley, "Metabolic function and properties of 4-hydroxyphenylacetic acid 1-hydroxylase from Pseudomonas acidovorans," Journal of Bacteriology, 121(1): p. 272–285.
- 13 J. G. Bergmann and John. Sanik, "Determination of trace amounts of chlorine in naphtha," Analytical Chemistry, 1957. 29 (2) : p. 241–243.
- 14 C. K. Fulton and R. A. Cooper, "Catabolism of sulfamate by Mycobacterium sp," CF1. Environmental microbiology, 2005. 7(3): p. 378–381.
- 15 W. G. Weisburg, S. M. Barns, D. A. Pelletier, and D. J. Lane, "16S ribosomal DNA amplification for phylogenetic study," Journal of Bacteriology, 1991. 173(2): p. 697–703.
- 16 S. Kumar, G. Stecher, and K. Tamura, "MEGA7: Molecular Evolutionary Genetics Analysis version 7.0 for bigger datasets," Molecular Biology and Evolution, 2016. 33(7): p. 1870-1874.
- 17 J. Holt, N. Krieg, P. Sneath, and J. Staley, 1994, Baltimore, MD : Bergey's manual of determinative bacteriology Williams and Wilkins, p. 527.

- 18 U. Nübel, F. Garcia-Pichel, and G. Muyzer, "PCR primers to amplify 16S rRNA genes from cyanobacteria," Applied and Environmental Microbiology, 1997. 63(8): p. 3327–3332.
- 19 P. J. Piggot and D. W. Hilbert, "Sporulation of Bacillus subtilis," Current Opinion in Microbiology, 2004. 7(6) :p. 579–586.
- 20 I. Tiago, I. Teixeira, S. Si1va, P. Chung, A. Veríssimo, and C. M. Manaia, Metabolic and genetic diversity of mesophilic and thermophilic bacteria isolated from composted municipal sludge on polyε-caprolactones, Current Microbiology, 2004. 49(6) : p. 407-414.
- 21 X.-H. Chen et al., "Difficidin and bacilysin produced by plant-associated Bacillus amyloliquefaciens are efficient in controlling fire blight disease," J Biotechnology, 2009. 140(1) : p. 38–44.
- V. Leclère et al., "Mycosubtilin overproduction by Bacillus subtilis BBG100 enhances the organism's antagonistic and biocontrol activities," Applied and Environmental Microbiology, 2005. 71(8) : p. 4577–4584.
- 23 R. Kotan, N. Dikbas, and H. Bostan, "Biological control of post harvest disease caused by Aspergillus flavus on stored lemon fruits," African Journal of Biotechnology, 2009. 8(2). p. 209-214.
- A. Lamine Senghor, W.-J. Liang, and W.-C. Ho, "Integrated control of Collectorichum gloeosporioides on mango fruit in Taiwan by the combination of Bacillus subtilis and fruit bagging," Biocontrol Science and Technology, 2007. 17(8): p. 865–870.
- 25 Z. Yang, H. Guo, and X. Zhang, "Study on the control of peach post-harvest diseases using Bacillus subtilis," China Fruits,2008. 23: p. 35–38.
- 26 D. Fravel, "Commercialization and implementation of biocontrol 1," Annual Review of Phytopathology, 43: p. 337–359.
- 27 S. Hughes, 1988, Microbial growth on 3-chloropropionic acid. University of Wales Institute of Science and Technology (UWIST).